



This open access document is posted as a preprint in the Beilstein Archives at <https://doi.org/10.3762/bxiv.2025.19.v1> and is considered to be an early communication for feedback before peer review. Before citing this document, please check if a final, peer-reviewed version has been published.

This document is not formatted, has not undergone copyediting or typesetting, and may contain errors, unsubstantiated scientific claims or preliminary data.

Preprint Title A Convergent Synthetic Approach to the Tetracyclic Core Framework of Khayanolide-Type Limonoids

Authors Zhiyang Zhang, Jialei Hu, Hanfeng Ding, Li Zhang and Peirong Rao

Publication Date 19 März 2025

Article Type Full Research Paper

Supporting Information File 1 supporting information.pdf; 6.7 MB

ORCID® IDs Hanfeng Ding - <https://orcid.org/0000-0002-1781-4604>; Peirong Rao - <https://orcid.org/0009-0003-7405-9531>



License and Terms: This document is copyright 2025 the Author(s); licensee Beilstein-Institut.

This is an open access work under the terms of the Creative Commons Attribution License (<https://creativecommons.org/licenses/by/4.0>). Please note that the reuse, redistribution and reproduction in particular requires that the author(s) and source are credited and that individual graphics may be subject to special legal provisions.

The license is subject to the Beilstein Archives terms and conditions: <https://www.beilstein-archives.org/xiv/terms>.

The definitive version of this work can be found at <https://doi.org/10.3762/bxiv.2025.19.v1>

A Convergent Synthetic Approach to the Tetracyclic Core Framework of Khayanolide-Type Limonoids

Zhiyang Zhang¹, Jialei Hu^{2,3}, Hanfeng Ding^{1,2}, Li Zhang*¹ and Peirong Rao*²

Address: ¹Department of Chemistry, Zhejiang Sci-Tech University, Hangzhou 310018, China. ²Department of Chemistry, Zhejiang University, Hangzhou 310058, China and ³Current address: Hangzhou DAC Biotechnology Co., Ltd 369 Qiaoxin Road, Qiantang District, Hangzhou 310018, Zhejiang, China.

Email: Peirong Rao - prao@zju.edu.cn, Li Zhang - lizhang@zstu.edu.cn

* Corresponding author

Abstract

A convergent approach for the enantioselective construction of an advanced intermediate containing the [5,5,6,6] tetracyclic core framework of the khayanolide-type limonoids was described. The strategy features an acylative kinetic resolution of the benzylic alcohol, a 1,2-Grignard addition and an AcOH-interrupted Nazarov cyclization.

Keywords

khayanolide-type limonoids; enantioselective synthesis; tetracyclic framework; interrupted Nazarov cyclization

Introduction

Limonoids, a class of tetranortriterpenoids derived biosynthetically from oxidative truncation of apotirucallane or apoeuphane precursors coupled with subsequent β -furan annulation [1–6], constitute an architecturally sophisticated family of natural products. These molecules exhibit a remarkable pharmacological portfolio encompassing anticancer, antimicrobial, anti-inflammatory, and insect antifeedant activities [7–9], positioning them as compelling targets for both therapeutic development and agrochemical innovation. Furthermore, limonoids are also renowned for their extraordinary structural complexity. For instance, the phragmalin-type limonoids represent a significant category of highly oxygenated rearranged limonoids, which contain a distinctive octahydro-1H-2,4-methanoindene cage. The fascinating architectures and remarkable biological profiles of these compounds have attracted widespread attention from synthetic community. In 1989, Corey and Hahl made a seminal contribution [10] to the field by completing the total synthesis of azadiradione (**1**). Following this, Lay and his team successfully synthesized azadirachtin (**2**) in 2007, a limonoid extensively utilized in organic agriculture [11]. In recent years, remarkable progress has been made on the total syntheses of various limonoids, with notable contributions from researchers such as Williams [12], Yamashita [13], Hao/Yang/Shen [14], Gong/Hao/Yang [15], Newhouse [16–19], Yang/Chen [20], Renata [21], Qin/Yu [22], Ma [23], Watanabe [24] and Li [25]. Their groundbreaking work has provided valuable insights and inspired further research in synthetic chemistry involving limonoids.

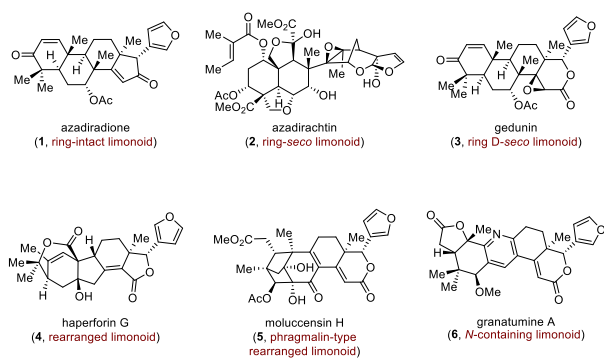
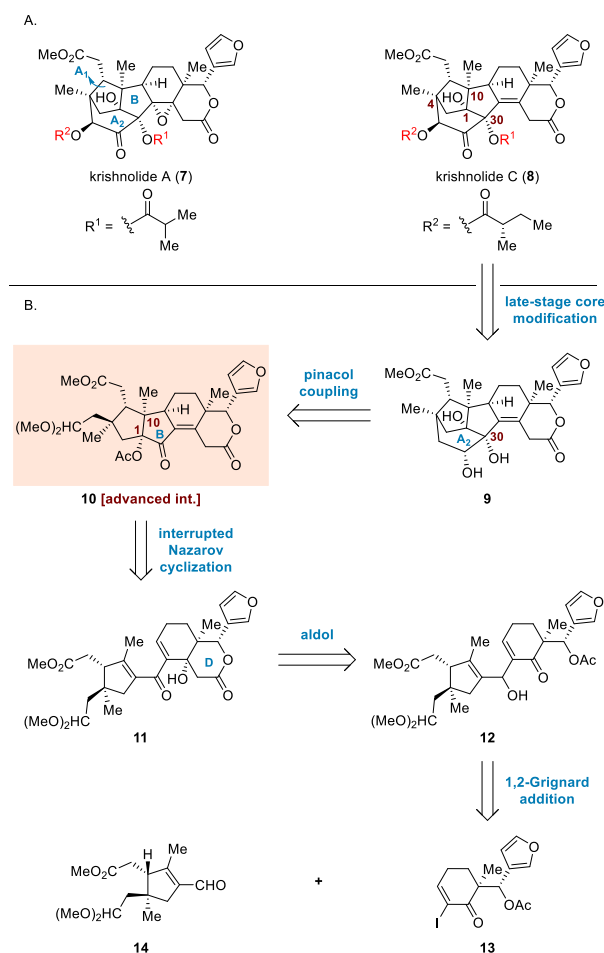


Figure 1: Representative Limonoid Triterpenes.

Krishnolides A and C (**7** and **8**, respectively; Scheme 1A) were identified by Wu and co-workers from the seeds of a Krishna mangrove *Xylocarpus moluccensis* [26]. These two molecules belong to khayanolides, a class of rearranged phragmalin limonoids characterized by a structurally intricate tricyclo[4.2.11^{0,30}.1^{1,4}]decane ring system. Additionally, krishnolides A and C contain 9–11 stereogenic centers and exhibit diverse oxidation patterns. Their relative and absolute configurations were determined through NMR, HR-ESIMS and ECD experiments, as well as single crystal X-ray diffraction analysis. Preliminary investigation revealed that krishnolide A (**7**) exhibited unique anti-Human Immunodeficiency Virus (HIV) activity, making the first report of anti-HIV activity in khayanolide-type limonoid. However, the highly oxygenated and polycyclic scaffolds pose substantial challenges toward their total synthesis. Two synthetic studies were disclosed successively by Sarpong [27] and Jirgensons [28], both focusing on construction of the unique methanoindene cage structure (A₁A₂B ring system). Building upon our previous syntheses of phragmalin-type limonoids [29], we herein disclose a convergent approach leveraging an AcOH-interrupted Nazarov cyclization to establish the [5,5,6,6] tetracyclic scaffold with precise stereochemical fidelity.

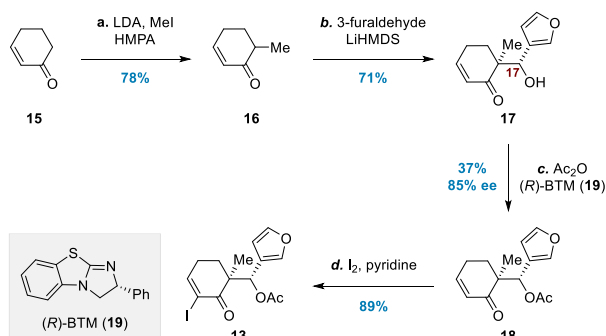
Results and Discussion

Our retrosynthetic analysis toward kishnolides A (**7**) and C (**8**) is delineated in Scheme 1B. We hypothesized that these two molecules could be synthesized from diol **9** through a late-stage modification involving adjustment of the oxidation state and regioselective acylation. The formation of **9** was envisioned to proceed via an intramolecular pinacol coupling [30,31] of [5,5,6,6] tetracycle **10**, which forged the A₂ ring while simultaneously installing the hydroxyl group at C30. The latter intermediate could in turn be derived from dienone **11** by an AcOH-interrupted Nazarov cyclization [32–34], thereby establishing the B ring with the desired all-*cis* stereochemical configuration, including the quaternary carbon at C10 and the essential tertiary alcohol at C1. The β-hydroxy lactone moiety (D ring) in **11** could be introduced through an intramolecular aldol condensation [35] of diketo-acetate **12**. Ultimately, the preparation of **12** could be traced back to aldehyde **14** through 1,2-Grignard addition with an organomagnesium reagent [36] prepared from α-iodoenone **13**.



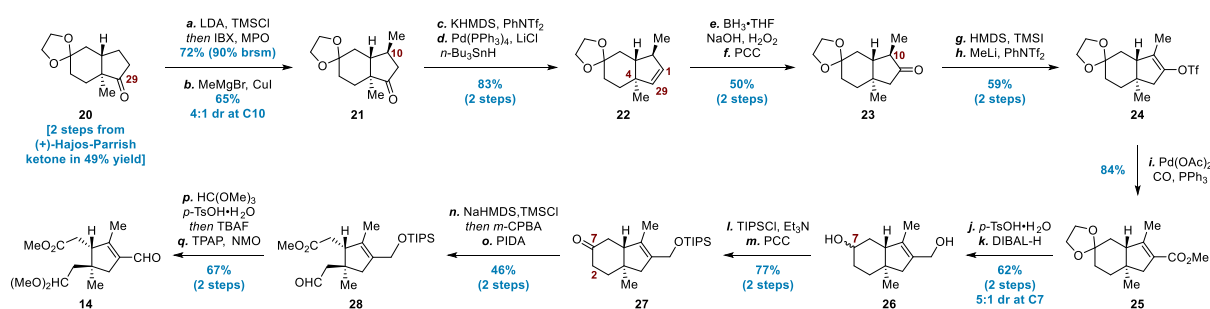
Scheme 1: Structures and Retrosynthetic Analysis of Krishnolides A (**7**) and C (**8**).

Our synthesis began with the preparation of α -iodoenone **13** (Scheme 2). The α -monomethylation of cyclohexenone **15** was efficiently carried out with LDA, HMPA and MeI [37], producing enone **16** in 78% yield. Subsequently, a diastereoselective aldol reaction between **16** and 3-furaldehyde promoted by LiHMDS gave alcohol **17** with high regioselectivity after an extensive screening of bases (LDA, NaHMDS, KHMDS, etc.). Drawing inspiration from the pioneering work of Birman [38], as well as Newhouse's applications [18,19], an acylative kinetic resolution of the benzylic alcohol was achieved by using (R)-BTM (**19**), furnishing acetate **18** with satisfactory efficiency and enantioselectivity (37% yield, 85% ee). Finally, iodination of **18** employing Johnsen's protocol (I_2 , pyridine) [39] provided α -iodoenone **13** in 89% yield.



Scheme 2: Construction of α -Iodoenone **13**.

On the other hand, the synthesis of acetaldehyde **14** commenced with bicyclic ketone **20**, which was prepared from (+)-Hajos–Parrish ketone in 49% yield over two steps (Scheme 3) [40–43]. Ensuring silyl enol etherification of the ketone at C29 coupled with IBX-mediated Nicolaou oxidation [44] furnished the corresponding enone in 72% yield (90% brsm). The methyl group at C10 was then introduced via a Michael addition (MeMgBr, CuI) to afford **21** in a yield of 65% (4:1 dr at C10). Initial attempts on the carbonyl 1,2-transposition protocol reported by Dong and co-workers were ineffective [45], leading to premature hydride termination and the formation of alkene **22**. As an alternative solution, by treating **21** with KHMDS and PhNTf₂, enol triflation took place successfully. The resultant triflate was coupled with *n*-Bu₃SnH to afford $\Delta^{1,29}$ -alkene **22** in 83% yield over two steps. Subsequent hydroboration-oxidation by employing BH₃·THF proceeded smoothly, providing a 4.4:1 mixture of regioisomeric alcohols with the desired isomer being the major component, presumably due to the considerable steric hindrance from the quaternary carbon at C4. However, decagram-scale separation of these two isomers by chromatography proved troublesome. Fortunately, the distinct reactivity of those two alcohols toward oxidation allowed for the selective conversion of desired alcohol to ketone **23** using PCC, producing a 50% overall yield, while the recovered undesired alcohol could be reverted to **21** by Swern oxidation.

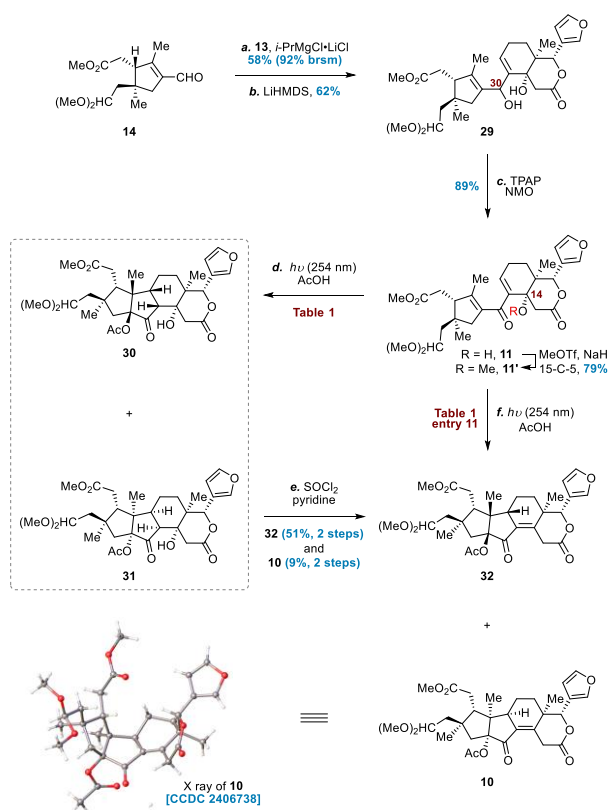


Scheme 3: Construction of Aldehyde **14**.

Direct enol triflation ($\text{Et}_3\text{N}/\text{Tf}_2\text{O}$, $\text{NaH}/\text{PhNTf}_2$, $\text{DTBMP}/\text{Tf}_2\text{O}$, etc.) of **23** led only to epimerization at C10 or slow decomposition of the starting material (see Supporting Information File 1, Table S1). Pleasingly, treating **23** with HMDS and TMSI regioselectively generated the expected TMS enol ether [46], which underwent a Li/I exchange in the presence of MeLi followed by interception of the lithium enolate with PhNTf₂ to give enol triflate **24** in 59% overall yield. The following palladium-catalyzed methoxycarbonylation produced methyl ester **25** in a satisfactory yield of 84%. TIPS-protected allylic alcohol **27** was selected as the appropriate precursor for α,β -unsaturated aldehyde and synthesized from **25** via a four-step transformation sequence, including deketalization, reduction, regioselective silylation of the primary alcohol and oxidation of the secondary alcohol. For the disconnection of C2–C7 bond, a two-step protocol involving Rubottom oxidation and PIDA-promoted oxidative cleavage [47] was applied to deliver aldehyde **28**. Finally, a one-pot acetalization and desilylation effectively afforded the acetal alcohol, which was then oxidized with the aid of TPAP, furnishing acetal aldehyde **14** in 67% yield over two steps.

With the two fragments **13** and **14** in hand, the next stage was set for the convergent construction of the target skeleton through a 1,2-addition (Scheme 4). Preliminary trials to generate organometallic species via Li/I exchange under various conditions (*n*-BuLi, *t*-BuLi, or *t*-BuLi in combination with CeCl_3 or MgBr_2) led to rapid decomposition, likely due to inherent instability of α -iodoenone **13**. Inspired by the influential studies by

Knochel [36] and Baran [48], we discovered that Mg/I exchange of **13** could be accomplished with *i*-PrMgCl·LiCl at $-78\text{ }^{\circ}\text{C}$. The resulting Grignard reagent reacted smoothly with aldehyde **14**, affording the corresponding adduct **12** in 58% yield (92% brsm). Since the newly created configuration at C30 was inconsequential, an intramolecular aldol reaction was directly carried out by treatment with LiHMDS to furnish β -hydroxy lactone **29**, which could be converted to dienone **11** through TPAP oxidation.



Scheme 4: Synthesis of the Advanced Intermediate **10**.

Having secured **11**, we proceeded to evaluate the pivotal Nazarov cyclization under a variety of conditions (Table 1). Initial trials under acid-mediated Nazarov conditions (AlCl₃, BF₃·Et₂O and Me₂AlCl) led to complete decomposition, while the exposure to AcOH resulted in recovery of the starting material (entries 1–4). Recognizing the limitations of these approaches, we then turned to the milder photo-Nazarov cyclization, in which the UV-light sources were found to be critical. While irradiation at

365 or 313 nm failed to induce cyclization and only allowed recovery of the starting material (entries 6 and 7), the disrotatory cyclization of **11** in the presence of AcOH by exposure to UV-light at 254 nm occurred exclusively to provide an inseparable mixture of **30** and **31** (entry 5). Subsequent dehydration of the resultant mixture with SOCl₂ and pyridine yielded separable enones **32** (57%) and **10** (7%) over two steps. The structure of **10** was unequivocally determined through X-ray crystallographic analysis (ORTEP drawing, Scheme 4). Further optimization by elevating the reaction temperature did not noticeably alter the ratio of **30** and **31** (entries 8 and 9). Moreover, attempts to apply interrupted Nazarov cyclization with H₂O under neutral condition merely resulted in decomposition (entry 10). To our delight, photo-irradiation of the corresponding methyl ether **11'** at 254 nm in the presence of AcOH at 20 °C led to a smooth cyclization, followed by spontaneous elimination, which produced the desired **10** as the single product, with no detection of **32** (entry 11). This result presumably arises from the minimization of dipole–dipole repulsions between the carbonyl of dienone moiety and the C14-OMe within the desired transition state.

Table 1: Optimization of the Interrupted Nazarov Cyclization^a.

entry	conditions	Yield (%) ^b	
		32	10
1 ^c	11 , AlCl ₃ , DCE, 60 °C	0	0
2 ^c	11 , BF ₃ ·Et ₂ O, DCE, 60 °C	0	0
3 ^c	11 , Me ₂ AlCl, toluene, 100 °C	0	0
4	11 , AcOH, DCE, 70 °C	0	0
5	11 , 254 nm <i>hν</i> , AcOH, DCM, 20 °C	57	7
6 ^d	11 , 313 nm <i>hν</i> , AcOH, DCM, 20 °C	0	0
7 ^d	11 , 365 nm <i>hν</i> , AcOH, DCM, 20 °C	0	0
8	11 , 254 nm <i>hν</i> , AcOH, DCE, 40 °C	54	8
9	11 , 254 nm <i>hν</i> , AcOH, DCE, 70 °C	51	9
10 ^c	11 , 254 nm <i>hν</i> , H ₂ O, DCM, 20 °C	0	0
11	11' , 254 nm <i>hν</i> , AcOH, DCM, 20 °C	0 ^e	73 ^e

^aReaction conditions: substrate (0.015 mmol), acid (2.0 equiv), solvent (5.0 mL).

^bIsolated yields over two steps involving Nazarov cyclization and dehydration.

^cDecomposition. ^dNo reaction. ^eIsolated yields over Nazarov cyclization/elimination cascade.

Conclusion

In conclusion, we have developed a convergent approach for the enantioselective assembly of an advanced intermediate en route to kishnolides A and C. Key steps of our strategy entail an acylative kinetic resolution of the benzylic alcohol, a 1,2-Grignard addition and an AcOH-interrupted Nazarov cyclization. Further elaboration of **10** to kishnolides A and C, as well as other khayanolide-type limonoids is currently ongoing, and the results will be disclosed in future reports.

Supporting Information

Deposition Number 2406738 contains the supplementary crystallographic data for this paper. These data can be obtained free of charge via the joint Cambridge Crystallographic Data Centre (CCDC) and Fachinformationszentrum Karlsruhe Access Structures service.

Supporting Information File 1:

File Name: supporting information

File Format: pdf

Title: Experimental procedures, NMR spectra and other characterization data for all new compounds.

Acknowledgements

We thank Mr. Jiyong Liu from the Chemistry Instrumentation Center Zhejiang University for X-ray crystallographic analysis.

Funding

Financial support was provided by the National Natural Science Foundation of China (22125109, 22371250 and 22401251).

References

1. Heasley, B. *Eur. J. Org. Chem.* **2011**, 2011, 19–46. doi:10.1002/ejoc.201001218
2. Tan, Q. G.; Luo, X. D. *Chem. Rev.* **2011**, 111, 7437–7522. doi:10.1021/cr9004023
3. Gualdani, R.; Cavalluzzi, M. M.; Lentini, G.; Habtemariam, S. *Molecules* **2016**, 21, 1530. doi:10.3390/molecules21111530
4. Zhang Y.; Xu, H. *RSC Adv.* **2017**, 7, 35191–35220. doi:10.1039/c7ra04715k
5. Fu, S.; Liu, B. *Org. Chem. Front.* **2020**, 7, 1903–1947. doi:10.1039/d0qo00203h
6. Luo, J.; Sun, Y.; Li, Q.; Kong, L. *Nat. Prod. Rep.* **2022**, 39, 1325–1365. doi:10.1039/d2np00015f
7. Champagne, D. E.; Koul, O.; Isman, M. B.; Scudder, G. G. E.; Towers, G. H. N. *Phytochemistry* **1992**, 31, 377–394. doi:10.1016/0031-9422(92)90003-9
8. Jain, D. C.; Tripathi, A. K. *Phytother. Res.* **1993**, 7, 327–334. doi:10.1002/ptr.2650070502
9. Roy, A.; Saraf, S. *Biol. Pharm. Bull.* **2006**, 29, 191–201. doi:10.1248/bpb.29.191
10. Corey, E. J.; Hahl, R. W. *Tetrahedron Lett.* **1989**, 30, 3023–3026. doi:10.1016/s0040-4039(00)99392-4
11. Veitch, G. E.; Beckmann, E.; Burke, B. J.; Boyer, A.; Maslen, S. L.; Ley, S. V. *Angew. Chem. Int. Ed.* **2007**, 46, 7629–7632. doi:10.1002/anie.200703027
12. Faber, J. M.; Eger, W. A.; Williams, C. M. *J. Org. Chem.* **2012**, 77, 8913–8921. doi:10.1021/jo301182f

13. Yamashita, S.; Naruko, A.; Nakazawa, Y.; Zhao, L.; Hayashi, Y.; Hiramata, M. *Angew. Chem. Int. Ed.* **2015**, *54*, 8538–8541. doi:10.1002/anie.201503794
14. Lv, C.; Yan, X. H.; Tu, Q.; Di, Y. T.; Yuan, C. M.; Fang, X.; Ben-David, Y.; Xia, L.; Gong, J. X.; Shen, Y. M.; Yang, Z.; Hao, X. J. *Angew. Chem. Int. Ed.* **2016**, *55*, 7539–7543. doi:10.1002/anie.201602783
15. Lv, C.; Tu, Q.; Gong, J. X.; Hao, X. J.; Yang, Z. *Tetrahedron* **2017**, *73*, 3612–3621. doi:10.1016/j.tet.2017.03.072
16. Schuppe, A. W.; Newhouse, T. R. *J. Am. Chem. Soc.* **2017**, *139*, 631–634. doi:10.1021/jacs.6b12268
17. Schuppe, A. W.; Huang, D.; Chen, Y.; Newhouse, T. R. *J. Am. Chem. Soc.* **2018**, *140*, 2062–2066. doi:10.1021/jacs.7b13189
18. Schuppe, A. W.; Zhao, Y.; Liu, Y.; Newhouse, T. R. *J. Am. Chem. Soc.* **2019**, *141*, 9191–9196. doi:10.1021/jacs.9b04508
19. Schuppe, A. W.; Liu, Y.; Gonzalez-Hurtado, E.; Zhao, Y.; Jiang, X.; Ibarra, S.; Huang, D.; Wang, X.; Lee, J.; Loria, J. P.; Dixit, V. D.; Li, X.; Newhouse, T. R. *Chem* **2022**, *8*, 2856–2887. doi:10.1016/j.chempr.2022.09.012
20. Zhang, W.; Zhang, Z. Y.; Tang, J. C.; Che, J. T.; Zhang, H. Y.; Chen, J. H.; Yang, Z. *J. Am. Chem. Soc.* **2020**, *142*, 19487–19492. doi:10.1021/jacs.0c10122
21. Li, J.; Chen, F.; Renata, H. *J. Am. Chem. Soc.* **2022**, *144*, 19238–19242. doi:10.1021/jacs.2c09048
22. Deng, H. P.; Deng, H. J.; Kim, C.; Li, P.; Wang, X. D.; Yu, Y. H.; Qin, T. *Nat. Synth.* **2023**, *3*, 378–385. doi:10.1038/s44160-023-00437-w
23. Jin, S. C.; Ma, D. W. *ChemRxiv*. **2024**, DOI: 10.26434/chemrxiv-2024-dscn9.
24. Mori, N.; Kitahara, T.; Mori, K.; Watanabe, H. *Angew. Chem. Int. Ed.* **2015**, *54*, 14920–14923. doi:10.1002/anie.201507935

25. Liu, X. T.; Xu, Y. Y.; Li, L. L.; Li, J. *J. Am. Chem. Soc.* **2024**, *146*, 26243–26250. doi:10.1021/jacs.4c07956
26. Zhang, Q.; Satyanandamurty, T.; Shen, L.; Wu, J. *Mar. Drugs* **2017**, *15*, 333. doi:10.3390/md15110333
27. Lebold, T. P.; Gallego, G. M.; Marth, C. J.; Sarpong, R. *Org. Lett.* **2012**, *14*, 2110–2113. doi:10.1021/ol300647k
28. Becica, J.; Rāciņš, O.; Ivanova, M.; Jirgensons, A. *J. Org. Chem.* **2023**, *88*, 10306–10309. doi:10.1021/acs.joc.3c00952
29. Rao, P. R.; Tang, D. M.; Xia, Q. D.; Hu, J. L.; Lin, X. F.; Xuan, J.; Ding, H. F. *J. Am. Chem. Soc.* **2025**, *147*, 4, 3003–3009. doi:10.1021/jacs.4c16265
30. Hovey, M. T.; Cohen, D. T.; Walden, D. M.; Cheong, P. H.-Y.; Scheidt, K. A. *Angew. Chem. Int. Ed.* **2017**, *56*, 9864–9867. doi:10.1002/anie.201705308
31. Sasano, Y.; Koyama, J.; Yoshikawa, K.; Kanoh, N.; Kwon, E.; Iwabuchi, Y. *Org. Lett.* **2018**, *20*, 3053–3056. doi:10.1021/acs.orglett.8b01087
32. Grant, T. N.; Rieder, C. J.; West, F. G. *Chem. Commun.* **2009**, 5676–5688. doi:10.1039/b908515g
33. Yadykov, A. V.; Shirinian, V. Z. *Adv. Synth. Catal.* **2020**, *362*, 702–723. doi:10.1002/adsc.201901001
34. Trudel, V.; Tien, C. H. *Nat. Rev. Chem.* **2021**, *5*, 604–623. doi:10.1038/s41570-021-00304-2
35. Fernandez-Mateos, A.; Benito, M. G.; Coca, G. P.; Gonzalez, R. R.; Hernandez, C. T. *Tetrahedron* **1981**, *51*, 7521–7526. doi:10.1016/0040-4020(95)00375-I
36. Ren, H.; Krasovskiy, A.; Knochel, P. *Org. Lett.* **2004**, *6*, 4215–4217. doi:10.1021/ol048363h

37. Marques, F. A.; Lenz, C. A.; Simonelli, F.; Noronha Sales Maia, B. H. L.; Vellasco, A. P.; Eberlin, M. N. *J. Nat. Prod.* **2004**, *67*, 1939–1941. doi:10.1021/np049771x
38. Birman, V. B.; Li, X. Benzotetramisole: *Org. Lett.* **2006**, *8*, 1351–1354. doi:10.1021/ol060065s
39. Johnson, C. R.; Adams, J. P.; Braun, M. P.; Senanayake, C. B. W.; Wovkulich, P. M.; Uskoković, M. R. *Tetrahedron Lett.* **1992**, *33*, 917–918. doi:10.1016/s0040-4039(00)91575-2
40. Daniewski, A. R.; Kiegiel, J. *Synth. Commun.* **1988**, *18*, 115–118. doi:10.1080/00397918808057826
41. Corey, E. J.; Huang, A. X. *J. Am. Chem. Soc.* **1999**, *121*, 710–714. doi:10.1021/ja983179a
42. Daniewski, A. R.; Liu, W. *J. Org. Chem.* **2001**, *66*, 626–628. doi:10.1021/jo0014414
43. Hamphreys, D. J.; Newall, C. E.; Paskins, H. A.; Phillips, G. H. *J. Chem. Soc., Perkin Trans. 1*, **1978**, 15–19. doi:10.1039/p19780000015
44. Nicolaou, K. C.; Gray, D. L. F.; Montagnon, T.; Harrison, S. T. *Angew. Chem. Int. Ed.* **2002**, *41*, 996–1000. doi:10.1002/1521-3757(20020315)114:6<1038::aid-ange1038>3.0.co;2-y
45. Wu, Z.; Xu, X. L.; Wang, J. C.; Dong, G. B. *Science* **2021**, *374*, 734–740. doi:10.1126/science.abl7854
46. Moher, E. D.; Collins, J. L.; Grieco, P. A. *J. Am. Chem. Soc.* **1992**, *114*, 2764–2765. doi:10.1021/ja00033a087
47. Ohno, M.; Oguri, I.; Eguchi, S. *J. Org. Chem.* **1999**, *64*, 8995–9000. doi:10.1021/jo990704v

Cernijenko, A.; Risgaard, R.; Baran, P. S. *J. Am. Chem. Soc.* **2016**, *138*, 9425–9428. doi:10.1021/jacs.6b06623