

Supporting Information

For

Norbornadiene functionalized triaza-triangulenium and trioxa-triangelium platforms

Roland Löw, Tobias Moje, Fynn Röhricht, Rainer Herges*

Address: Otto Diels Institute for Organic Chemistry, University of Kiel, Otto-Hahn-Platz 4, 24118 Kiel, Germany

Email: Rainer Herges - rherges@oc.uni-kiel.de

* Corresponding author

Analytical equipment and methods, experimental procedures and NMR spectra

Table of Contents

- I. Analytical equipment and methods
- II. Experimental procedures
- III. NMR spectra
- IV. UV-Vis absorption spectra
- V. Kinetic studies in solution by ^1H NMR spectroscopy
- VI. Calculations

I. Analytical equipment and methods

NMR Spectroscopy

NMR spectra were measured in deuterated solvents (Deutero). All compounds were characterized using ¹H and ¹³C NMR spectroscopy. The signals were assigned using 2D spectroscopy. For ¹H and ¹³C NMR assignment we performed HSQC and HMBC experiments. The degree of deuteration is given in parentheses. ¹H NMR spectra are referenced to the following signals:

chloroform-d (99.8%): $\delta = 7.26$ ppm. (s)

benzene-d₆ (99.8%): $\delta = 7.16$ ppm. (s)

acetone-d₆ (99.5%): $\delta = 2.05$ ppm. (quint.)

The signal multiplicities are abbreviated as follows:

s: singlet, d: doublet, t: triplet, m: multiplet, dt: double triplet, ps. t: pseudo triplet, dd: double doublet, td: triple doublet.

Measurements were performed by the following instruments:

Bruker CABAV 500neo (¹H NMR: 500 MHz, ¹³C NMR: 125 MHz, ¹¹B NMR: 160 MHz,

¹⁹F NMR: 470 MHz, ²⁹Si NMR: 99 MHz)

Bruker AV 600 (¹H NMR: 600 MHz, ¹³C NMR: 150 MHz)

IR spectroscopy

Infrared spectra were measured on a Perkin-Elmer 1600 Series FT-IR spectrometer with an A531-G Golden-Gate-Diamond-ATR-unit. Signals were abbreviated with w, m, s and for weak, medium and strong intensities. Broad signals are additionally labeled with br.

Mass spectrometry

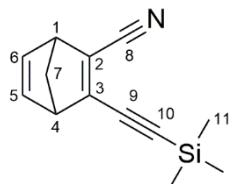
The high resolution (HR) mass spectra were measured with an APEX 3 FT-ICR with a 7.05 T magnet by co. Bruker Daltonics. Electron impact (EI). Electrospray ionization (ESI) mass spectra were measured with a Thermo Scientific Q EXACTIVE.

Chromatography stationary phases

For column chromatography purifications silica gel (Merck, particle size 0.040–0.063 mm) was used. R_f values were determined by thin layer chromatography on Polygram® Sil G/UV254 (Macherey-Nagel, 0.2 mm particle size).

II. Experimental procedures

II.1 3-[2-Trimethylsilyl-ethynyl]-bicyclo[2.2.1]hepta-2,5-diene-2-carbonitrile (5).



In toluene (24 mL), 3-bromo-bicyclo[2.2.1]hepta-2,5-diene-2-carbonitrile **4**^[1] (600 mg, 3.06 mmol) was dissolved under nitrogen atmosphere, trimethylsilylacetylene (522 μ L, 3.67 mmol), Pd(PPh₃)₄ (106 mg, 91.8 μ mol), copper(I)iodide (58.3 mg, 306 μ mol) and triethylamine (1.06 mL, 7.65 mmol) were added and the mixture was stirred for 80 min at 60 °C. The mixture was filtered through celite and the solvent was removed under reduced pressure. The crude product was purified via column chromatography (silica gel, cyclohexane/ ethyl acetate, 4/1) to obtain a yellow liquid (468 mg, 2.19 mmol, 72%).

¹H NMR (500.1 MHz, CDCl₃, 298 K, TMS): δ = 6.85-6.81 (m, 2H, H-5, H-6), 3.86-3.83 (m, 1H, H-1), 3.77-3.73 (m, 1H, H-4), 2.27 (dt, ³J = 7.0 Hz, ⁴J = 1.6 Hz, 1H, H-7_a), 2.18 (dt, ³J = 7.0 Hz, ⁴J = 1.6 Hz, 1H, H-7_b), 0.24 (s, 9H, H-11) ppm.

¹³C NMR (125.8 MHz, CDCl₃, 298 K, CHCl₃): δ = 154.09 (s, C-2), 142.02 (s, C-5), 141.49 (s, C-6), 129.79 (s, C-3), 115.04 (s, C-9), 97.63 (s, C-8), 73.10 (s, C-7), 57.32 (s, C-4), 54.19 (s, C-1), -0.22 (s, C-11) ppm.

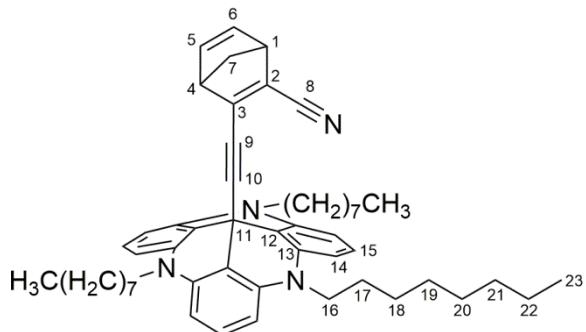
²⁹Si NMR (99.4 MHz, CDCl₃, 300 K, TMS): δ = -16.23 ppm.

MS (EI, 70eV): m/z = 213.1 [M]⁺.

IR: $\tilde{\nu}$ = 2927 (w), 2852 (w), 2207 (m), 2139 (w), 1576 (w), 1557 (w), 1450 (w), 1302 (m), 1251 (m), 1132 (w), 1068 (w), 1019 (w), 840 (vs), 760 (m), 733 (s), 702 (w), 626 (m), 534 (m) cm⁻¹.

HRMS (EI, 70 eV): m/z [M]⁺ calcd. for C₁₃H₁₅NSi: 213.09738, found: 213.09724.

II.2 12c-Bicyclo[2.2.1]hepta-2,5-diene-2-carbonitrile-ethynyl-4,8,12-tri-n-octyl-4,8,12-triazatriangulene (1).



In tetrahydrofuran (abs., 60 mL) 3-[2-trimethylsilyl-ethynyl]-bicyclo[2.2.1]hepta-2,5-diene-2-carbonitrile **5** (100 mg, 469 μ mol) was dissolved under nitrogen atmosphere, octyl-TATA-BF₄

6^[2] (397 mg, 562 µmol) and powdered potassium hydroxide (263 mg, 3.69 mmol) were added and the mixture was refluxed for 5 h. The mixture was poured onto saturated sodium chloride solution (50 mL) and the aqueous phase extracted with diethyl ether (3x 50 mL). The combined organic layers were dried over magnesium sulfate and the solvent was removed under reduced pressure. The crude product was purified via column chromatography (aluminium oxide basic, diethyl ether) and recrystallized from ethanol to obtain an orange solid (222 mg, 292 µmol, 62%).

¹H NMR (600.1 MHz, C₆D₆, 298 K, TMS): δ = 7.21 (t, ³J = 8.2 Hz, 3H, H-15), 6.61 (m, 6H, H-14), 5.91-5.86 (m, 1H, H-5), 5.76-5.73 (m, 1H, H-6), 3.86-3.80 (ps. t, 6H, H-16), 2.89-2.86 (m, 1H, H-4), 2.82-2.79 (m, 1H, H-1), 1.86-1.77 (m, 6H, H-17), 1.34-1.20 (m, 32H, H-7_a, H-7_b, H-18, H-19, H-20, H-21, H-22), 0.94-0.90 (ps. t, 9H, H-23) ppm.

¹³C NMR (150.9 MHz, C₆D₆, 298 K, TMS): δ = 153.78 (s, C-2), 141.34 (s, C-5), 141.11 (s C-13), 140.83 (s, C-6), 129.37 (s, C-3), 129.09 (s, C-15), 109.44 (s, C-12), 105.55 (s, C-14), 79.48 (s, C-10), 72.23 (s, C), 56.36 (s, C-1), 53.57 (s, C-4), 47.08 (s, C-16), 32.22 (s, C), 30.17 (s, C-11), 29.76 (s, C), 29.69 (s, C-7), 27.24 (s, C), 25.93 (s, C-17), 23.09 (s, C-18), 14.40 (s, C-22) ppm.

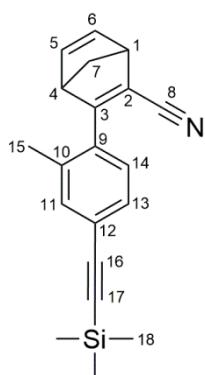
MS (MALDI-TOF): m/z = 759.1 [M]⁺.

IR (ATR): $\tilde{\nu}$ = 2953 (m), 2922 (m), 2851 (m), 2207 (w), 1617 (s), 1579 (vs), 1481 (s), 1456 (vs), 1394 (vs), 1372 (m), 1267 (m), 1246 (m), 1207 (w), 1167 (s), 1147 (m), 908 (w), 766 (vs), 731 (vs), 657 (w), 637 (m), 609 (w) cm⁻¹.

m.p. = 101.7 °C.

Elemental analysis calcd. (%) for C₅₃H₆₆N₄: C 83.86; H 8.76; N 7.38; found: C 83.538; H 8.649; N 7.322.

II.3 3-[2-methyl-4-trimethylsilyl]ethynylphenyl]-bicyclo[2.2.1]hepta-2,5-diene-2-carbonitrile (10).



In a solution of toluene (15 mL), ethanol (3.75 ml) and H₂O (750 µL) 2-[2-methyl-4-[2-(trimethylsilyl)ethynyl]phenyl]-4,4,5,5-tetramethyl-1,3,2-dioxaborolane **9^[3]** (402 mg, 1.28 mmol), 3-bromo-bicyclo[2.2.1]hepta-2,5-diene-2-carbonitrile **4** (250 mg, 1.28 mmol), Pd(PPh₃)₄ (73.9 mg, 64.0 µmol) and sodiumcarbonate (339 mg, 3.20 mmol) were suspended under nitrogen atmosphere and refluxed for 19 h. To the mixture H₂O (10 mL) was added and the layers were separated. The water layer was extracted with dichloromethane (3x 30 mL) and the combined organic layers were dried over magnesium sulfate. The solvent

was removed under reduced pressure and the crude product was purified via column chromatography (silica gel, cyclohexane/ethyl acetate, 1/1) to obtain a yellow oil (59.0 mg, 194 μ mol, 38%).

$^1\text{H NMR}$ (500.1 MHz, acetone-d₆, 298 K, TMS): δ = 7.39 (s, 1H, H-11), 7.33 (dd, 3J = 8.0 Hz, 4J = 1.0 Hz, 1H, H-13), 7.19 (d, 3J = 8.0 Hz, 1H, H-14), 7.11-7.04 (m, 2H, H-5, H-6), 3.99-3.96 (m, 2H, H-1, H-4), 2.45 (td, 3J = 6.9 Hz, 4J = 1.6 Hz, 1H, H-7_a), 2.32 (s, 3H, H-9), 2.21 (td, 3J = 6.9 Hz, 4J = 1.6 Hz, 1H, H-7_b), 0.24 (s, 9H, H-18) ppm.

$^{13}\text{C NMR}$ (125.8 MHz, acetone-d₆, 298 K, TMS): δ = 174.06 (s, C-3), 143.99 (s, C), 142.56 (s, C), 136.95 (s, C-10), 135.87 (s, C-9), 134.81 (s, C-11), 130.03 (s, C-13), 128.20 (s, C-14), 124.68 (s, C-12), 123.58 (s, C-2), 105.54 (s, C-16), 95.69 (s, C-17), 73.68 (s, C-7), 58.28 (s, C), 55.42 (s, C), 20.36 (s, C-9), 0.00 (s, C-18) ppm.

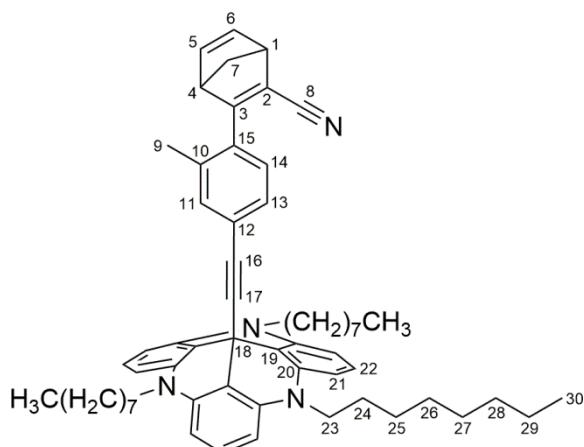
$^{29}\text{Si NMR}$ (99.4 MHz, acetone-d₆, 298 K, TMS): δ = -17.48 ppm.

MS (EI, 70eV): m/z = 303.14 [M]⁺.

IR: $\tilde{\nu}$ = 2958 (br, w), 2204 (m), 2151 (w), 1606 (w), 1560 (w), 1493 (w), 1450 (w), 1310 (w), 1295 (m), 1233 (w), 1004 (w), 949 (w), 899 (w), 834 (vs), 814 (s), 759 (m), 723 (vs), 658 (m) cm⁻¹.

HRMS (EI, 70 eV): m/z [M]⁺ calcd. for C₂₀H₂₁NSi: 303.14433, found: 303.14410.

II.4 12c-(4-(Bicyclo[2.2.1]hepta-2,5-diene-2-carbonitrile)3-methylphenyl)ethynyl-4,8,12-tri-n-octyl-4,8,12-triazatriangulene (2).



In tetrahydrofuran (abs., 40 mL) 3-[2-methyl-4-trimethylsilylethynylphenyl]-bicyclo[2.2.1]hepta-2,5-diene-2-carbonitrile **10** (65.0 mg, 214 μ mol) was dissolved under nitrogen atmosphere and octyl-TATA-BF₄ **6** (181 mg, 257 μ mol) and powdered potassium hydroxide (95.9 mg, 1.71 mmol) were added and the mixture was refluxed for 1 h. The mixture was poured onto sat. sodium chloride solution (30 mL) and extracted with diethylether (3x 50 mL). The combined organic layers were dried over magnesium sulfate and the solvent was removed under reduced pressure. The crude product was purified via column chromatography (aluminium oxide basic, diethyl ether) and recrystallized from ethanol to obtain a grey solid (80.0 mg, 94.2 μ mol, 44%).

$^1\text{H NMR}$ (500.1 MHz, C₆D₆, 298 K, TMS): δ = 7.25 (t, 3J = 8.3 Hz, 3H, H-22), 6.85 (dd, 3J = 8.1 Hz, 4J = 1.1 Hz, 1H, H-13), 6.82 (s, 1H, H-11), 6.66-6.61 (m, 7H, H-21, H-14), 6.33 (dd, 3J = 5.1 Hz, 3J = 3.0 Hz, 1H, H-5), 6.19 (dd, 3J = 5.1 Hz, 3J = 3.0 Hz, 1H, H-6), 3.84-3.78 (ps. t, 6H, H-23), 3.24-3.21 (m, 1H, H-4), 3.02-2.99 (m, 1H, H-1), 1.85-1.77 (m, 6H, H-24), 1.76 (s, 3H, H-9), 1.64 (td, 3J = 6.8 Hz, 4J = 1.5 Hz, 1H, H-7_a), 1.54 (td, 3J = 6.8 Hz, 4J = 1.5 Hz, 1H, H-7_b), 1.31-1.15 (m, 30H, H-25, H-26, H-27, H-28, H-29), 0.91 (ps. t, 9H, H-30) ppm.

¹³C NMR (125.8 MHz, C₆D₆, 298 K, TMS): δ = 172.34 (s, C-3), 142.91 (s, C-5), 141.17 (s, C-20), 140.99 (s, C-6), 135.03 (s, C-10), 134.48 (s, C-11), 133.48 (s, C-15), 129.42 (s, C-13), 128.69 (s, C-22) 126.68 (s, C-14), 125.00 (s, C-12), 122.16 (s, C-2), 111.07 (s, C-19), 105.67 (s, C-21), 95.63 (s, C-16), 84.11 (s, C-17), 72.37 (s, C-7), 57.20 (s, C-1), 54.46 (s, C-4), 46.71 (s, C-23), 32.21 (s, C-27), 29.74 (s, C-28), 29.69 (s, C-29), 29.08 (s, C-18), 27.23 (s, C-25), 26.19 (s, C-24), 23.04 (s, C-26), 19.96 (s, C-9), 14.37 (s, C-30) ppm.

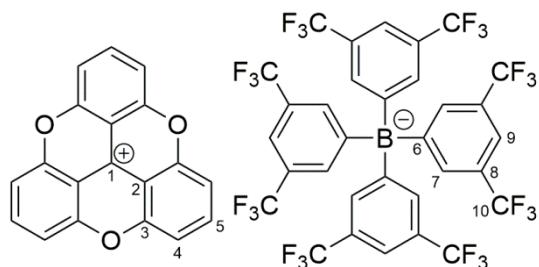
MS (MALDI-TOF): m/z = 849.4 [M]⁺.

IR: $\tilde{\nu}$ = 2922 (s), 2852 (m), 2204 (w), 1615 (s), 1579 (vs), 1482 (vs), 1456 (vs), 1393 (cs), 1373 (m), 1293 (w), 1267 (m), 1244 (m), 1167 (s), 1022 (w), 911 (w), 886 (w), 828 (w), 816 (w), 789 (w), 772 (m), 748 (m), 724 (s), 696 (vs), 657 (w), 608 (w) cm⁻¹.

m.p. = 73.6 °C.

Elemental analysis calcd. (%) for C₆₀H₇₂N₄: C 84.86; H 8.55; N 6.60; found: C 84.634; H 8.476; N 6.571.

II.5 Synthesis of 4,8,12-Trioxatrianguleniumtetrakis[3,5-bis-(trifluoromethyl)phenyl]borate (8)



In dichloro methane (200 mL) 4,8,12-trioxatrianguleniumtetrafluoroborate **7**^[4] (636 mg, 1.71 mmol) and sodium tetrakis[3,5-bis(trifluoromethyl)phenyl]borate (1.89 g, 2.10 mmol) were suspended and stirred at room temperature for 2 h. The mixture was filtered and the solution was washed with water (3x 150 mL) and dried over magnesium sulfate. The solvent was removed under reduced pressure and the crude product was dissolved in 25 mL ethyl acetate and precipitated by adding 400 mL cyclohexane. Filtration gave 1.77 g (1.54 mmol, 91%) of a yellowish solid.

¹H NMR (500.1 MHz, acetone-d₆, 298 K, TMS): δ = 8.66 (t, ³J = 8.5 Hz, 3H, H-5), 7.99 (d, ³J = 8.5 Hz, 6H, H-4), 7.79 (t, ⁴J = 2.5 Hz, 8H, H-7), 7.67 (s, 4H, H-9) ppm.

¹³C NMR (125.8 MHz, acetone-d₆, 298 K, TMS): δ = 162.6 (q, C-6), 154.7 (s, C-3), 144.7 (m, C-5), 135.5 (m, C-7), 130.0 (m, C-10), 125.4 (d, C-8), 118.4 (m, C-9), 113.6 (s, C-4), 107.3 (s, C-2) ppm.

¹⁹F NMR (470 MHz, aceton-d₆, 298 K, TMS): δ = -62.2 ppm.

¹¹B NMR (160 MHz, aceton-d₆, 298 K, TMS): δ = -5.86 ppm.

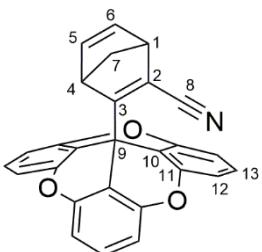
IR (ATR): $\tilde{\nu}$ = 2311 (w), 2164 (w), 1635 (s), 1552 (m), 1467 (m), 1355 (s), 1275 (s), 1143 (s), 1112 (s), 1063 (s), 1021 (s), 900 (m), 887 (m), 776 (s), 681 (s), 558 (s), 412 (m) cm⁻¹.

MS (ESI, pos): m/z = 285.05 [C₁₉H₉O₃]⁺.

MS (ESI, neg): m/z = 863.07 [C₃₂H₁₂BF₂₄]⁻.

m.p. = 202 °C.

II.6 12c-Bicyclo[2.2.1]hepta-2,5-diene-2-carbonitrile-4,8,12-trioxatriangulene (3).



In tetrahydrofuran (abs., 15 mL), 3-bromo-bicyclo[2.2.1]hepta-2,5-diene-2-carbonitrile **4** (213 mg, 1.09 mmol) was dissolved under nitrogen atmosphere und the solution was cooled to -78 °C. To the solution *n*-BuLi (436 µL, 1.09 mmol, 2.5 M in *n*-hexane) was added slowly and stirred for 45 min. 4,8,12-trioxatrianguleniumtetrakis-[3,5-bis(trifluormethyl)phenyl]borate **8** (1.38 g, 1.20 mmol), dissolved in tetrahydrofuran (abs., 30 mL), was added slowly and stirred for 45 min at -78 °C and further for 20 h at room temperature. To the solution, diethylether (30 mL) was added and the solution was washed with water (3x 50 mL). The combined organic layers were dried over magnesium sulfate and the solvent was removed under reduced pressure. The crude product was purified via column chromatography (alox basic, diethylether) and recrystallized from methanol to obtain a colorless solid (147 mg, 367 µmol, 42%).

¹H NMR (500.1 MHz, C₆D₆, 298 K, TMS): δ = 6.93 (t, ³J = 8.3 Hz, 3H, *H*-13), 6.87-6.81 (m, 6H, *H*-12), 6.01 (dd, ³J = 5.0 Hz, ³J = 3.1 Hz, 1H, *H*-5), 5.83 (dd, ³J = 5.0 Hz, ³J = 3.1 Hz, 1H, *H*-6), 3.55-3.53 (m, 1H, *H*-4), 3.03-3.00 (m, 1H, *H*-1), 1.25-1.20 (m, 2H, *H*-7) ppm.

¹³C NMR (125.8 MHz, C₆D₆, 298 K, TMS): δ = 173.68 (s, C-3), 153.31 (d, C-11), 142.32 (s, C-5), 140.30 (s, C-6), 129.90 (s, C-13), 120.43 (s, C-2), 111.92 (d, C-12), 109.93 (s, C-10), 70.67 (s, C-7), 55.41 (s, C-1), 52.55 (s, C-4), 31.28 (s, C-9) ppm.

MS (EI, 70eV): m/z = 401.07 [M]⁺.

IR (ATR): $\tilde{\nu}$ = 2946 (w), 2202 (w), 1743 (w), 1612 (s), 1481 (m), 1456 (s), 1306 (w), 1260 (vs), 1065 (m), 1041 (m), 1010 (vs), 934 (w), 903 (m), 877 (m), 786 (m), 773 (m), 688 (m), 597 (w), 576 (w) cm⁻¹.

HRMS (EI, 70 eV): m/z [M]⁺ calcd. for C₂₇H₁₅NO₃: 401.10519, found: 401.10515.

III. NMR spectra

III.1 3-[2-Trimethylsilyl-ethynyl]-bicyclo[2.2.1]hepta-2,5-diene-2-carbonitrile (4).

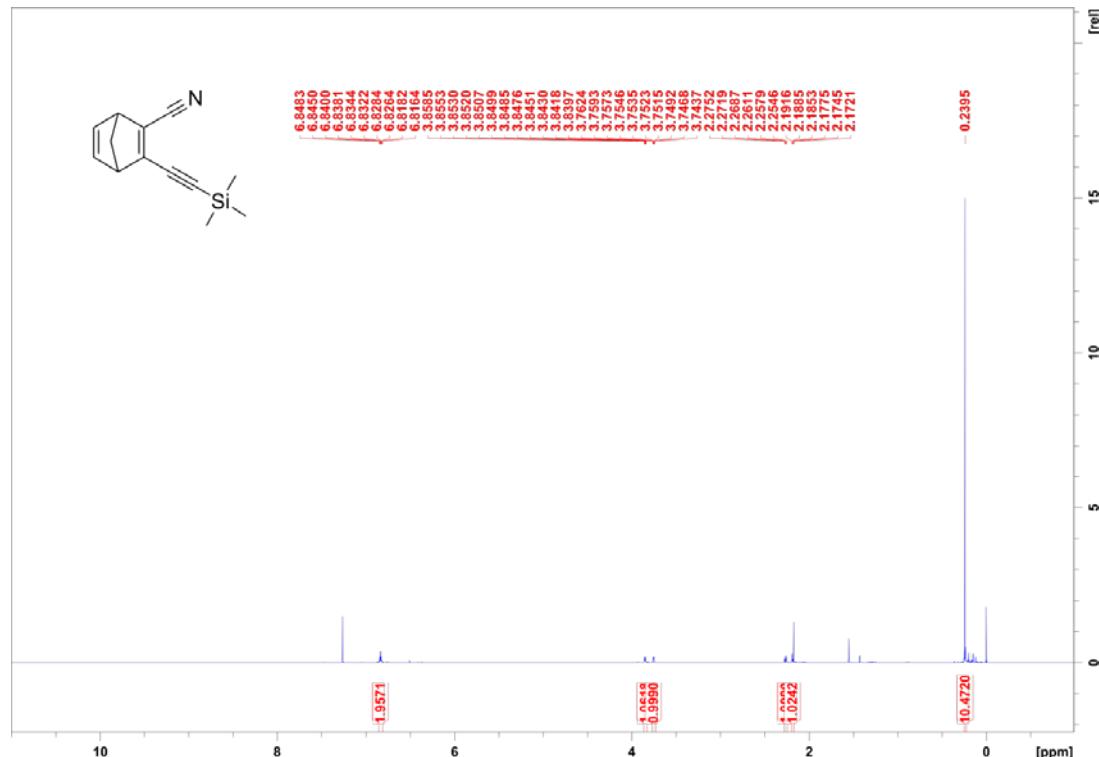


Figure S1. ^1H NMR spectrum (500.1 MHz, CDCl_3) of compound 5.

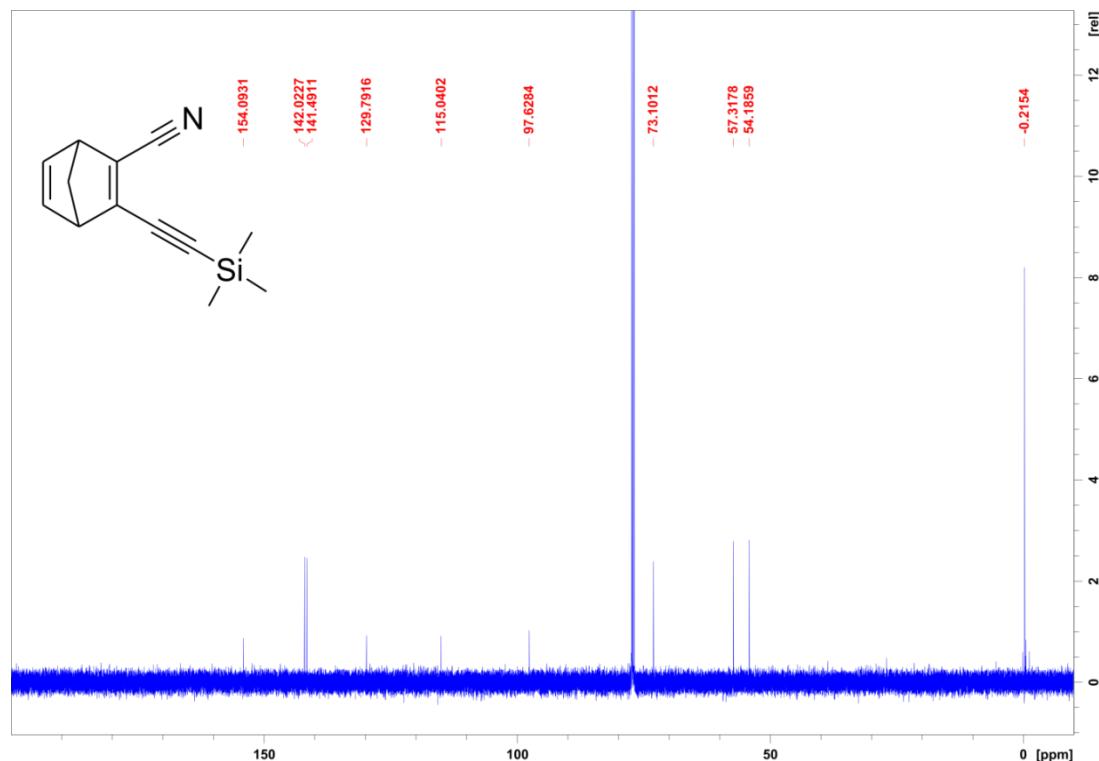


Figure S2. ^1H NMR spectrum (125.8 MHz, CDCl_3) of compound 5.

III.2 12c-Bicyclo[2.2.1]hepta-2,5-diene-2-carbonitrile-ethynyl-4,8,12-tri-*n*-octyl-4,8,12-triazatriangulene (1).

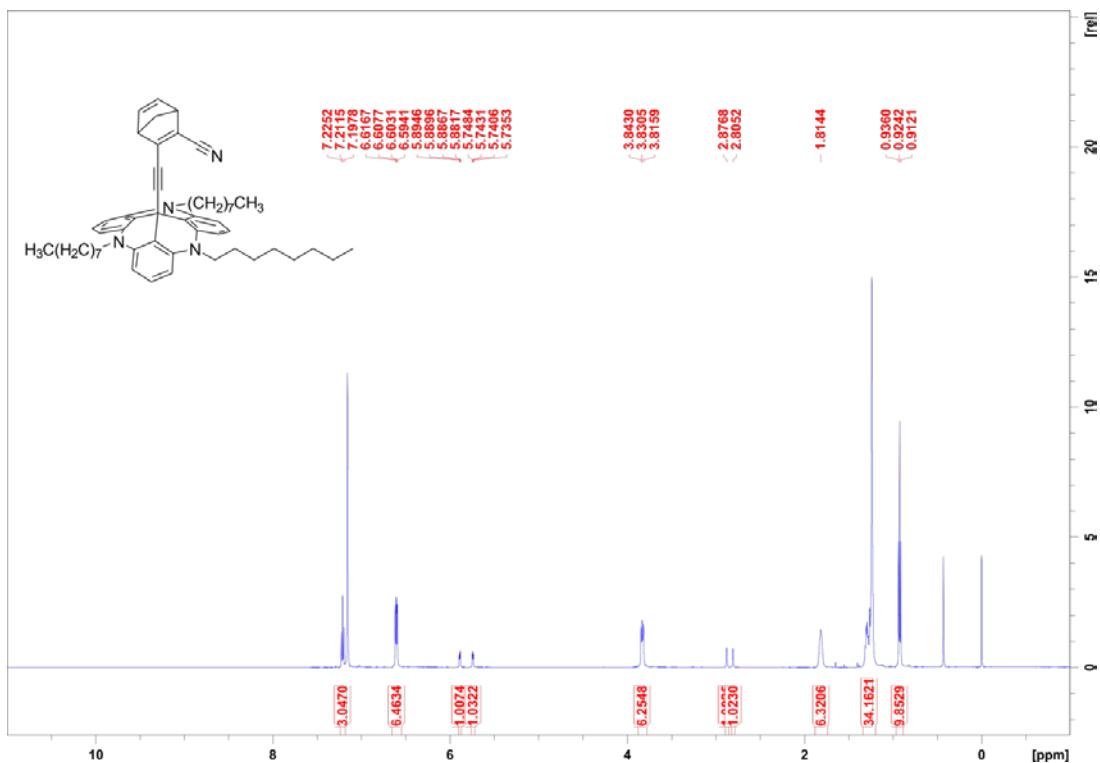


Figure S3. ^1H NMR spectrum (600.1 MHz, C_6D_6) of compound 1.

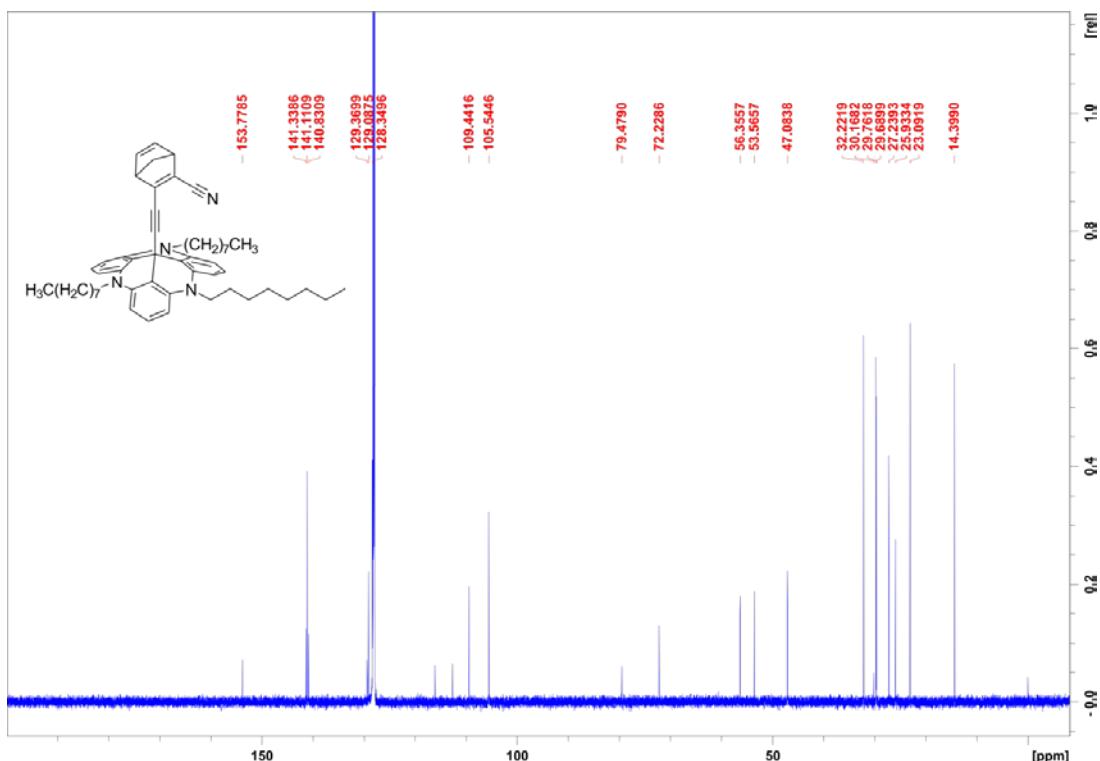


Figure S4. ^1H NMR spectrum (150.9 MHz, C_6D_6) of compound 1.

III.3 3-[2-Methyl-4-trimethylsilyl-ethynylphenyl]-bicyclo[2.2.1]hepta-2,5-diene-2-carbonitrile (10).

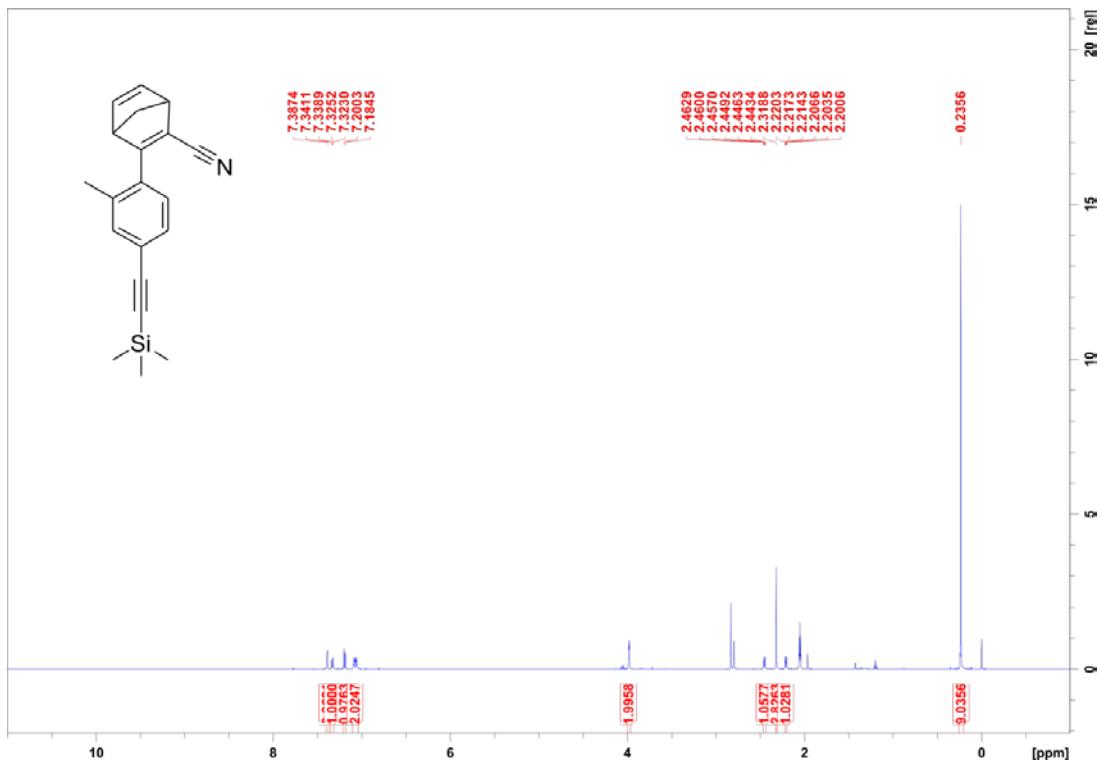


Figure S5. ^1H NMR spectrum (500.1 MHz, acetone- d_6) of compound 10.

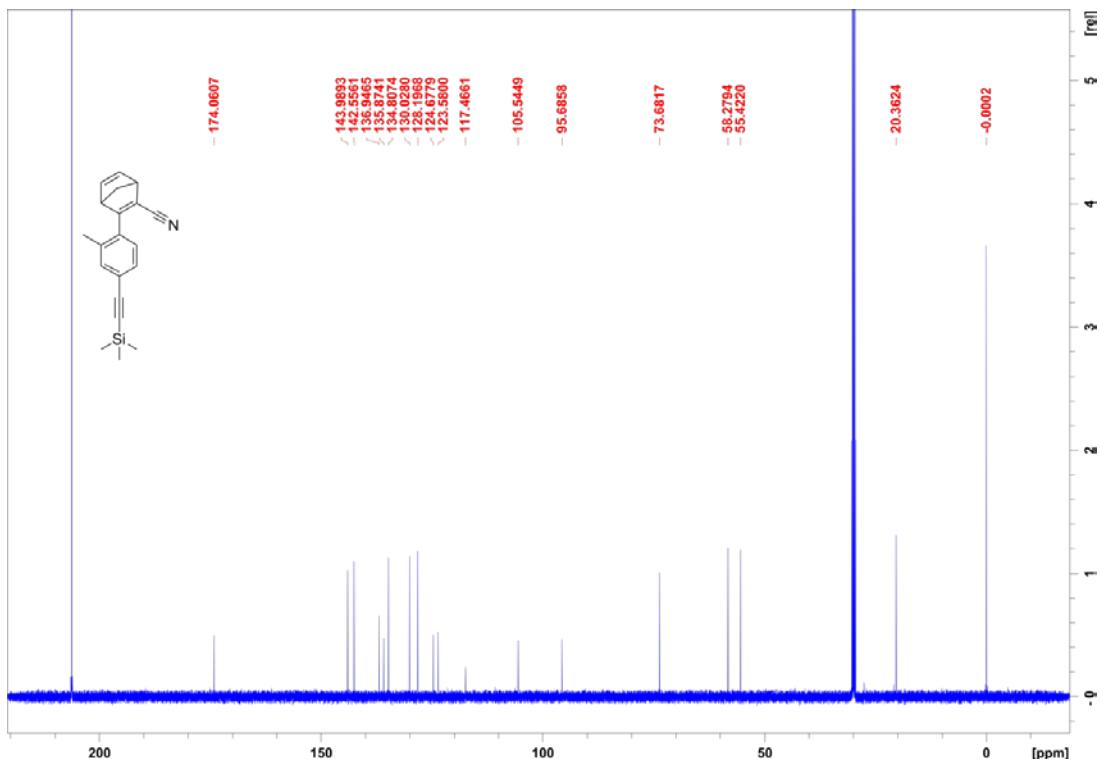


Figure S6. ^1H NMR spectrum (125.8 MHz, acetone- d_6) of compound 10.

III.4 12c-(4-(Bicyclo[2.2.1]hepta-2,5-diene-2-carbonitrile)3-methylphenyl)ethynyl-4,8,12-tri-*n*-octyl-4,8,12-triazatriangulene (2).

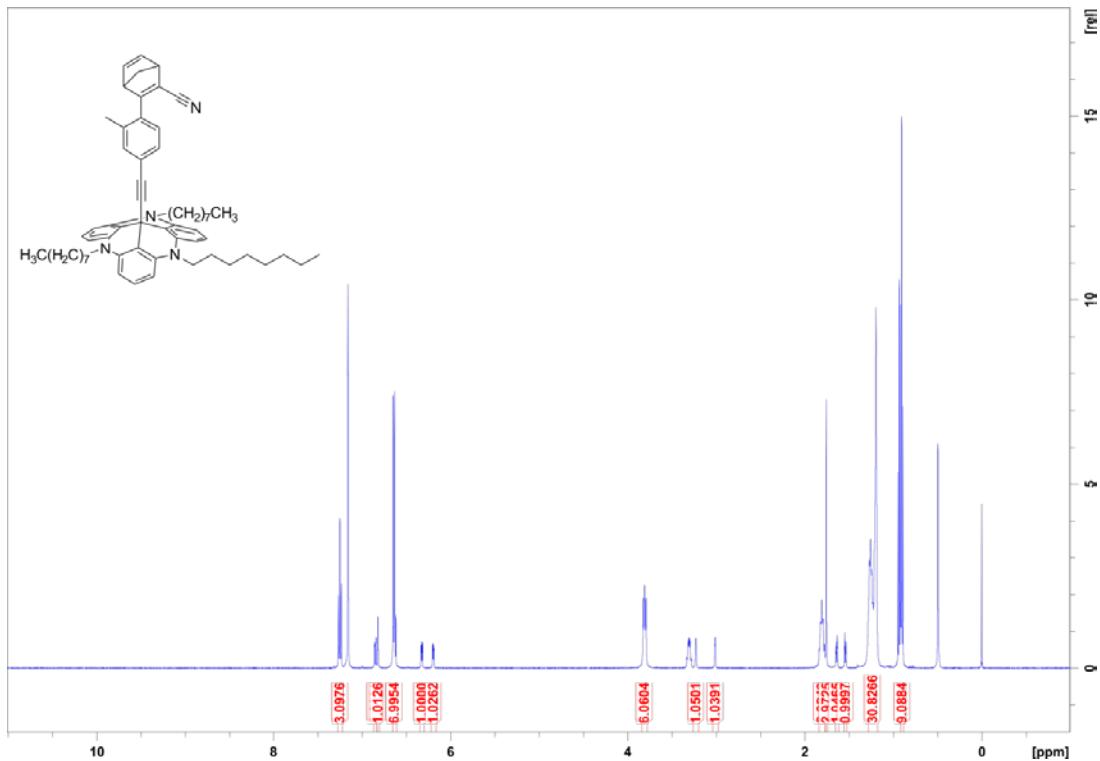


Figure S7. ^1H NMR spectrum (500.1 MHz, C_6D_6) of compound 2.

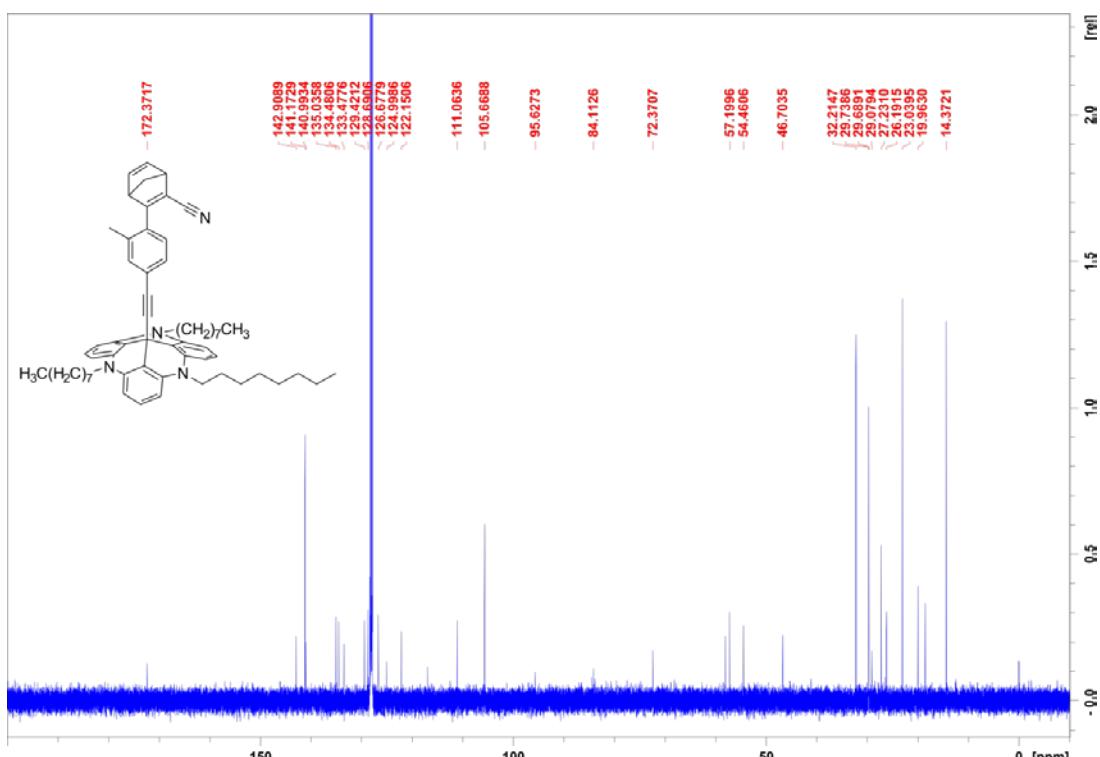


Figure S8. ^1H NMR spectrum (125.8 MHz, C_6D_6) of compound 2.

III.5 Synthesis of 4,8,12-Trioxatrianguleniumtetrakis[3,5-bis-(trifluoromethyl)phenyl]-borate (8)

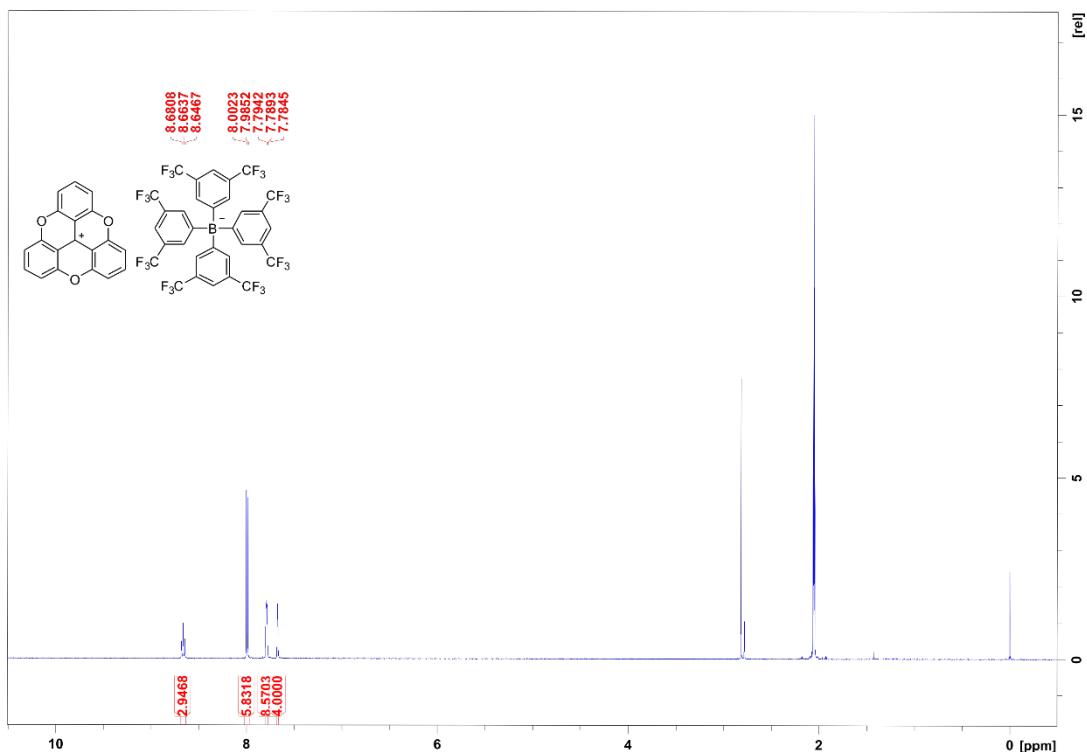


Figure S9. ¹H NMR spectrum (500.1 MHz, acetone-d₆) of compound 8.

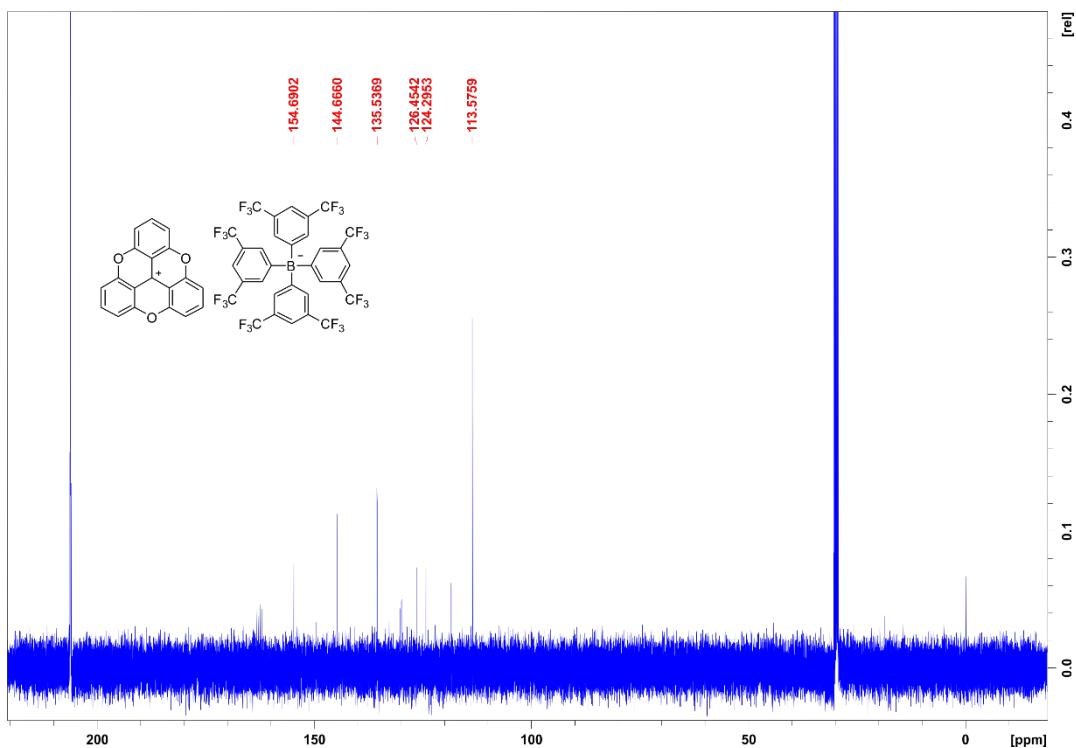


Figure S10. ¹³C NMR spectrum (125.8 MHz, acetone-d₆) of compound 8.

III.6 12c-Bicyclo[2.2.1]hepta-2,5-diene-2-carbonitrile-4,8,12-trioxatriangulene (3).

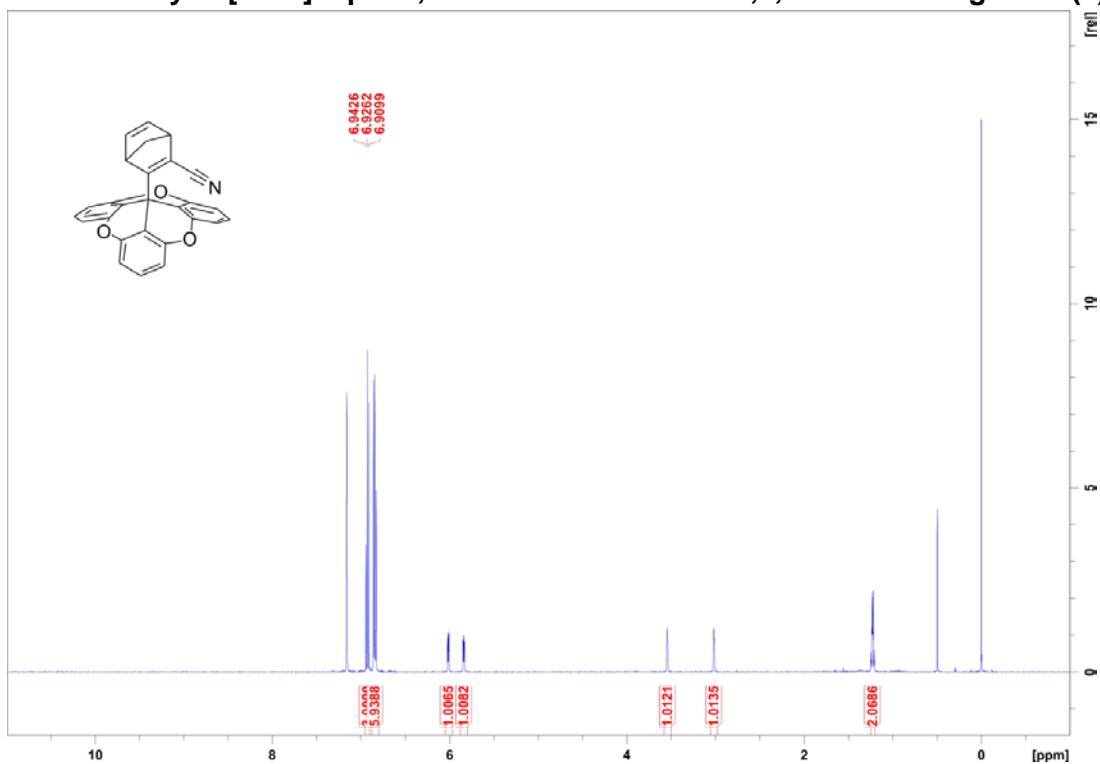


Figure S11. ¹H NMR spectrum (500.1 MHz, acetone-d₆) of compound 3.

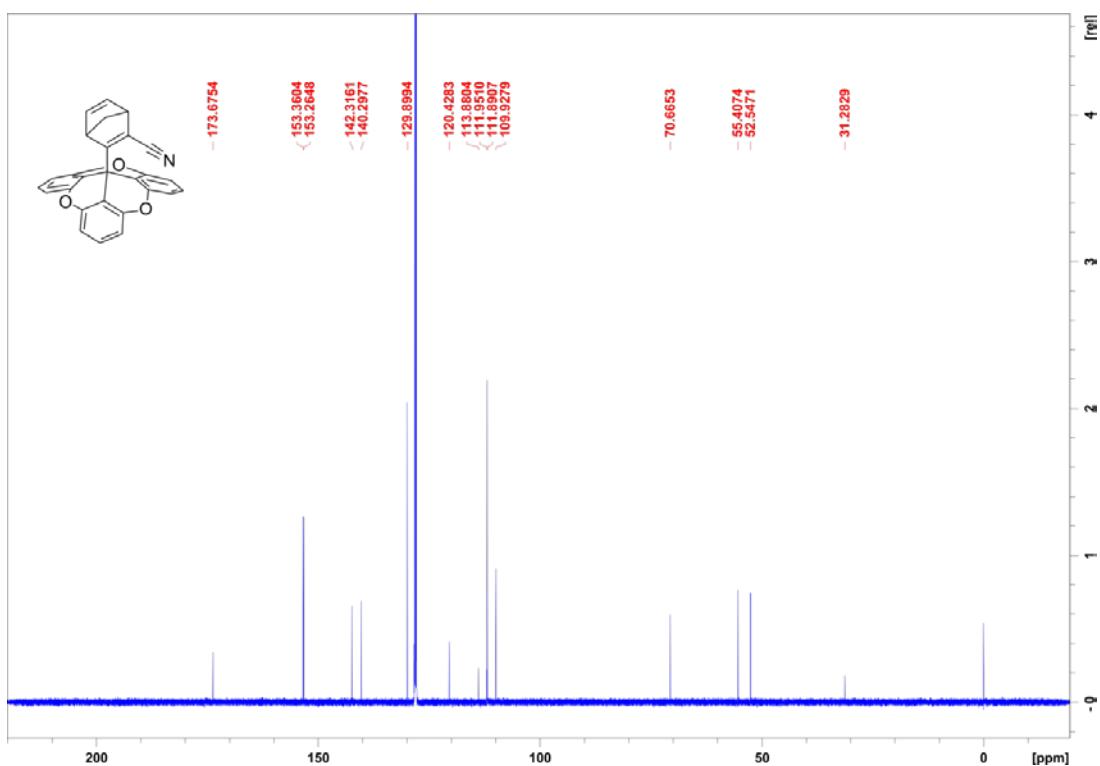


Figure S12. ¹H NMR spectrum (125.8 MHz, acetone-d₆) of compound 3.

IV. UV/Vis absorption spectra

IV.1 Methods

UV-Vis spectra were recorded on a PerkinElmer Lambda 650 Photospectrometer in a 1 cm path length quartz cuvette. Irradiation of UV/Vis samples were carried out at 25 °C using a self-built LED positioned at a distance of 1 cm from the sample.

IV.2 UV/Vis spectra

Compound 1:

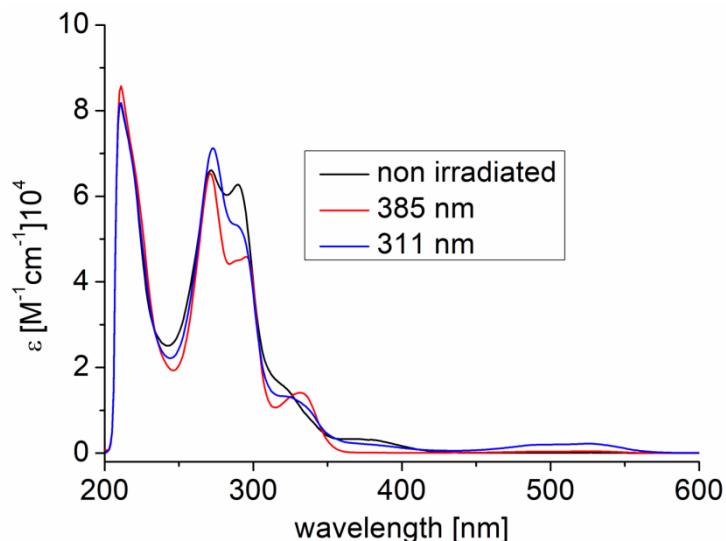


Figure S13. UV-Vis spectra of compound 1 in tetrahydrofuran at room temperature (31.6 $\mu\text{mol/L}$). Upon irradiation with 385 nm the [2+2] cycloaddition and with 311 nm the [2+2] cycloreversion take place with partly decomposition of 1.

Compound 2:

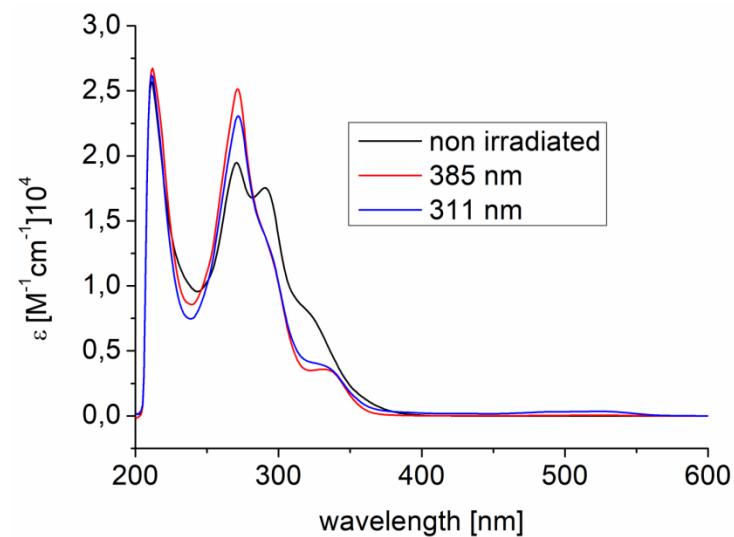


Figure S14. UV-Vis spectra of compound **2** in tetrahydrofuran at room temperature (30.6 $\mu\text{mol/L}$). Upon irradiation with 385 nm the [2+2] cycloaddition and with 311 nm the [2+2] cycloreversion take place with partly decomposition of **2**.

Compound 3:

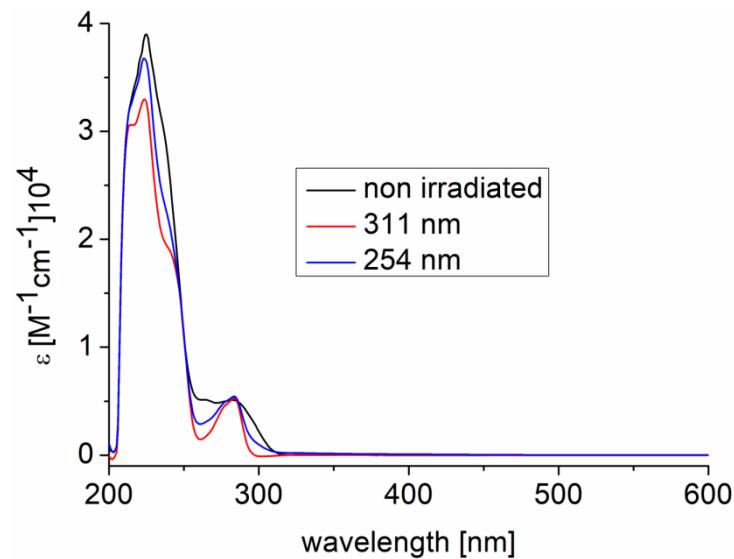


Figure S15. UV-Vis spectra of compound **3** in tetrahydrofuran at room temperature (79.7 $\mu\text{mol/L}$). Upon irradiation with 311 nm the [2+2] cycloaddition and with 254 nm the [2+2] cycloreversion take place.

V. Kinetic studies in solution by ^1H NMR spectroscopy

V.1 Thermal isomerization rate measurements by ^1H NMR

V.1.1 Compound 1: 12c-bicyclo[2.2.1]hepta-2,5-diene-2-carbonitrile-ethynyl-4,8,12-tri-*n*-octyl-4,8,12-triazatriangulene

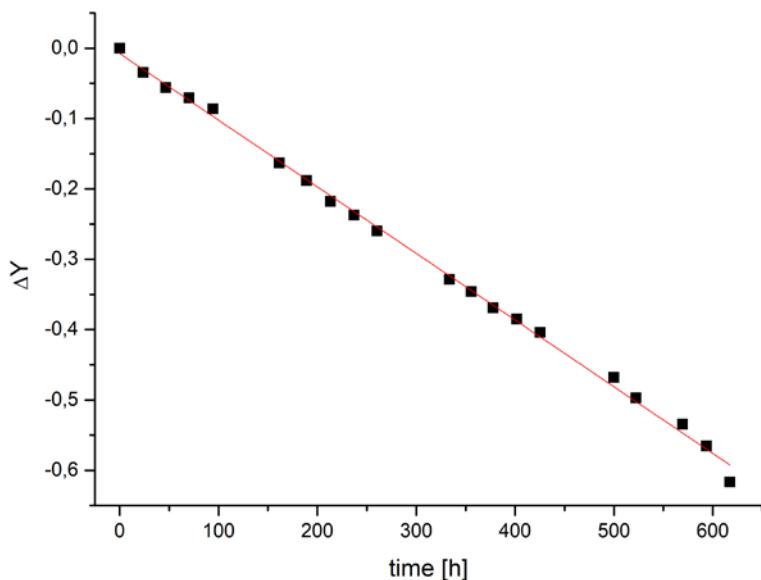


Figure S16. Determination of the thermal isomerization rate k of **1b** (QC) by ^1H NMR spectroscopy (toluene, 294.5 K, 800 $\mu\text{mol/L}$, under nitrogen). ΔY : $\ln \{ [\text{QC}]_t / [\text{QC}]_0 \}$, $[\text{QC}]_t$: ^1H NMR integral of the CH_2 group neighbouring the N bridge atom of the TATA platform in QC **1b** at time t , $[\text{QC}]_0$ corresponding ^1H integral at $t = 0$. A rate constant of $k = 0.95 \cdot 10^{-3} [\text{s}^{-1}]$ was determined from a linear fit of the $\Delta Y/t$ curve. The half-life of **1b** at 293.5 K in toluene was determined as 742.7 h.

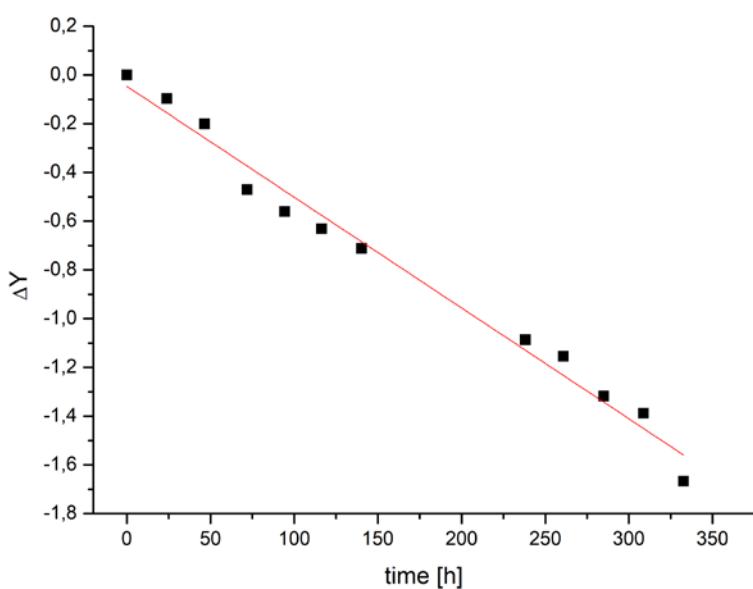


Figure S17. Determination of the thermal isomerization rate k of **1b** (QC) by ^1H NMR spectroscopy (toluene, 306 K, 800 $\mu\text{mol/L}$, under nitrogen). ΔY : $\ln \{ [\text{QC}]_t / [\text{QC}]_0 \}$, $[\text{QC}]_t$: ^1H NMR integral of the CH_2 group neighbouring the N bridge atom of the TATA platform in QC **1b** at time t , $[\text{QC}]_0$ corresponding ^1H integral at $t = 0$. A rate constant of $k = 4.55 \cdot 10^{-3} [\text{s}^{-1}]$ was determined from a linear fit of the $\Delta Y/t$ curve. The half-life of **1b** at 305 K in toluene was determined as 152.3 h.

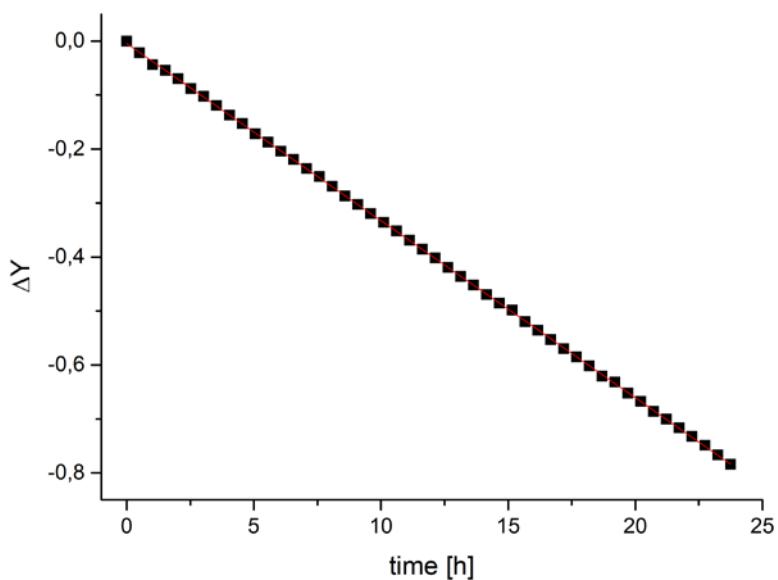


Figure S18. Determination of the thermal isomerization rate k of **1b** (QC) by ^1H NMR spectroscopy (toluene, 318 K, 800 $\mu\text{mol/L}$, under nitrogen). ΔY : $\ln \{ [\text{QC}]_t / [\text{QC}]_0 \}$, $[\text{QC}]_t$: ^1H NMR integral of the CH_2 group neighbouring the N bridge atom of the TATA platform in QC **1b** at time t , $[\text{QC}]_0$ corresponding ^1H integral at $t = 0$. A rate constant of $k = 3.28 \cdot 10^{-2} [\text{s}^{-1}]$ was determined from a linear fit of the $\Delta Y/t$ curve. The half-life of **1b** at 316 K in toluene was determined as 21.1 h.

V.2 Arrhenius Plots for compound 1 in solution

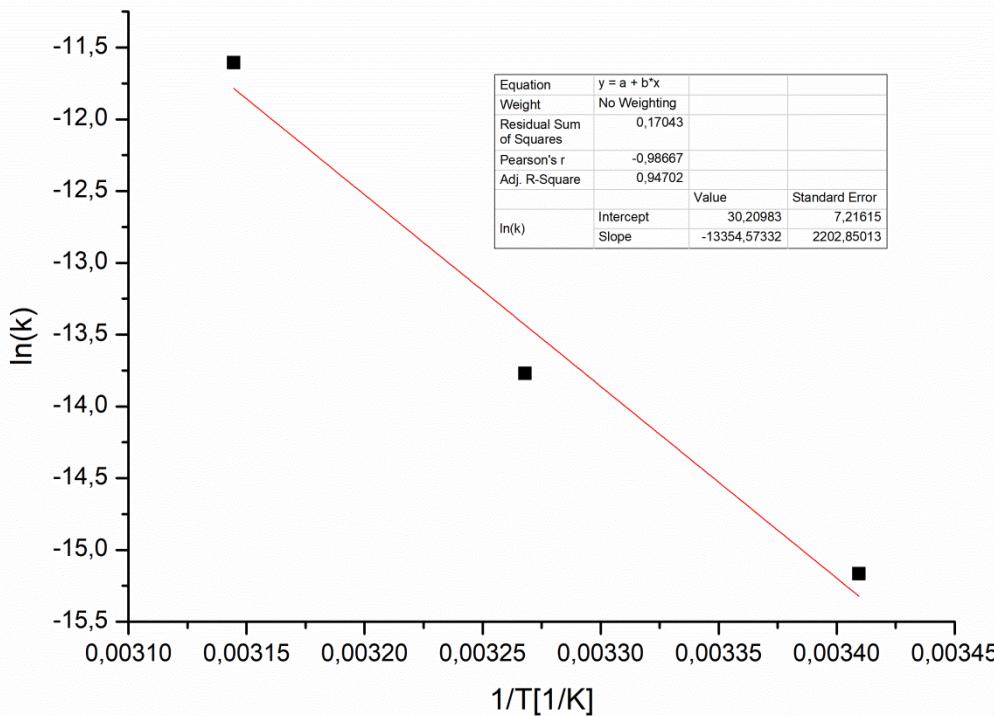


Figure S19. Arrhenius plot of the QC→NB isomerization of compound 1 which shows an activation energy of 111 KJ/mol.

VI. Calculations

General

All geometry optimizations were carried out using density functional theory with the Minnesota functional M06-2X^[5] in cooperation with Grimmes D3^[6] dispersion correction and the large triple zeta basis def2-TZVP.^[7] This level performed well in Grimme's study on basic properties of a selected data base of structures.^[8] The calculations were carried out with Turbomole7.2,^[9] the m4 grid (in Turbomole nomenclature) and resolution-of-identity (RI) with multipole accelerated RI-J (marij). All stationary points were characterized by frequency calculations.

Coordinates

1a Norbornadiene-ethinyl-TATA

$$E_{M062x-D3/def2TZVPP} = -1334.818694137$$

$$Nimag = 2 (-17.61 \text{ cm}^{-1}; -6.98 \text{ cm}^{-1})$$

C 0.2951221	-0.0928872	-0.1011107	H -1.9156187	-0.2236005	4.6214454	H -1.9634605	-2.0799268	3.0960594
C -1.1704890	0.0583325	0.0307701	H -1.8804372	4.2837995	-1.7919981	C 5.1902861	0.1449561	-0.3944998
C -1.8459803	-1.2871414	-0.0863013	H -2.5840991	0.8914577	-4.3295204	C 3.7945606	0.7798825	-0.2757764
C -1.4660755	0.6715255	1.3802029	H -2.4584959	3.3342520	-3.9952168	C 2.8939936	-0.2162297	-0.3304604
C -1.6505093	0.9771099	-1.0692086	C -2.2135302	-1.7702573	-1.3397407	C 3.6995631	-1.5188465	-0.4949560
C -2.0166023	0.4470018	-2.3024972	C -2.7092283	-3.0680069	-1.4664217	C 4.4984532	-1.7012154	0.8018948
C -2.3089692	1.2999839	-3.3661180	C -2.8470960	-3.8518472	-0.3315202	C 5.3851804	-0.7160310	0.8609508
C -2.2386146	2.6708262	-3.1696166	C -2.5305116	-3.3665867	0.9278284	C 4.8322591	-0.9762285	-1.4018125
C -1.9128988	3.2121107	-1.9358891	C -2.0351589	-2.0682900	1.0509749	H 3.1192811	-2.3681946	-0.8372177
C -1.6242241	2.3558268	-0.8734252	H -2.9850852	-3.4502756	-2.4404461	H 4.4734997	-0.5956125	-2.3564633
C -1.6581455	-0.1500244	2.4865432	H -3.2290342	-4.8592200	-0.4288256	H 5.6385924	-1.6930174	-1.5405454
C -1.4436158	2.0566046	1.5252156	H -2.6682762	-3.9798697	1.8085969	C 3.5404447	2.1563930	-0.0592153
C -1.5573038	2.6223135	2.7950065	N -1.7190682	-1.56269515	2.2918696	N 3.3664019	3.2799138	0.1124753
C -1.7095793	1.7905484	3.8931360	N -2.0723133	-0.9366409	-2.4432285	C 1.4937530	-0.1463250	-0.2039334
C -1.7761609	0.4122027	3.7570676	N -1.3062321	2.8403748	0.3878158	H 4.2997123	-2.4654899	1.5367172
H -1.5258145	3.6970613	2.9141541	H -1.3325165	3.8377176	0.5145511	H 6.0798673	-0.4901947	1.6549143
H -1.7950422	2.2281439	4.8787235	H -2.4407027	-1.2816335	-3.3135604	H 5.9899191	0.8322061	-0.6473880

1b Quadricyclane-ethinyl-TATA

$$E_{M062x-D3/def2TZVPP} = -1334.790048271$$

$$Nimag = 2 (-17.69 \text{ cm}^{-1}; -16.54 \text{ cm}^{-1})$$

C 0.3054503	0.3034247	-0.0945918	H -2.3276835	-1.0302136	-4.3946707	H -2.5118600	1.1332272	-3.3628401
C -1.1302770	-0.0443125	-0.0006913	H -1.0336213	-3.7179672	2.8571254	C 5.2248477	0.2524714	0.3025426
C -1.9801978	1.1980518	-0.1257588	H -2.0094006	0.0955646	4.5736198	C 3.9342807	-0.4616393	-0.0379931
C -1.4616538	-1.0110023	-1.1140504	H -1.5490045	-2.3171390	4.8222897	C 2.8957731	0.6715019	-0.2368182
C -1.3719566	-0.7026080	1.3381671	C -2.3001572	1.9364283	1.0109498	C 3.6884847	1.9334865	0.0194105
C -1.6973779	0.0753641	2.4446918	C -2.9682718	3.1546828	0.8838703	C 3.7715013	1.2663363	-1.3237662
C -1.7628444	-0.5075456	3.7096545	C -3.3237891	3.6014334	-0.3792977	C 4.7973406	0.1400815	-1.1295429
C -1.5059887	-1.8641665	3.8408492	C -3.0562099	2.8526103	-1.5147241	C 4.9144361	1.6269682	0.8459516
C -2.1252576	-2.6576889	2.7424688	C -2.3876446	1.6349595	-1.3835984	H 3.1346661	2.8545877	0.1149901
C -1.1582689	-2.0717652	1.4776801	H -3.2091368	3.7351609	1.7648138	H 4.6984026	1.6082974	1.9136120
C -1.8709972	-0.5278849	-2.3527081	H -3.8403526	4.5467600	-0.4793723	H 5.7259757	2.3295061	0.6545150
C -1.2454623	-2.3740050	-0.9257332	H -3.3640005	3.1992886	-2.4925361	C 3.6513980	-1.8297003	0.2286013
C -1.3857235	-3.2533699	-1.9996441	N -2.1129522	0.8365935	-2.4881291	N 3.4231702	-2.9358321	0.4371242
C -1.7589621	-2.7531021	-3.2371711	N -1.9403505	1.4332685	2.2557890	C 1.4871371	0.5111329	-0.1655574
C -2.0180939	-1.4040318	-3.4274294	N -0.8904824	-2.8190303	0.3395901	H 6.0527536	-0.3436335	0.6547414
H -1.2041283	-4.3108295	-1.8608406	H -0.7586756	-3.8087064	0.4598547	H 5.3362125	-0.3924080	-1.8947861
H -1.8658301	-3.4336704	-4.0711709	H -2.2768692	1.9408163	3.0566112	H 3.4516521	1.7002478	-2.2557280

2a Norbornadiene-me-phenyl-ethinyl-TATA

$$E_{M062x-D3/def2TZVPP} = -1605.175260624$$

$$Nimag = 2 (-14.49 \text{ cm}^{-1}; -5.84 \text{ cm}^{-1})$$

C -0.2457023	-1.4876625	0.6394580	C -2.7862046	-3.4233285	0.1223341	H -4.6567694	-3.8121724	-0.8663231
C -0.5345119	-2.8676838	1.0900366	C -0.5378466	-2.7990450	3.6024283	H -1.1383084	-5.2010861	-2.8979447
C 0.7645750	-3.6248130	1.2443716	C -2.6503133	-2.7136067	2.4365560	H -3.5872371	-4.9109227	-2.8011110
C -1.2622650	-2.8201868	2.4137094	C -3.3135472	-2.5177101	3.6480923	C 1.3142258	-4.2850416	0.1485604
C -1.4012621	-3.5439317	0.0519411	C -2.5748045	-2.4519134	4.8192594	C 2.5813256	-4.8598954	0.2459864
C -0.8075457	-4.2041110	-1.0202745	C -1.1970648	-2.6040793	4.8162472	C 3.2679201	-4.7802511	1.4476407
C -1.5980179	-4.6969740	-2.0579207	H -4.3912342	-2.4230832	3.6670754	C 2.7135300	-4.1658290	2.5597123
C -2.9730723	-4.5329856	-1.9949587	H -3.0871473	-2.2969734	5.7594210	C 1.4471049	-3.5902321	2.4571918
C -3.5806782	-3.9147659	-0.9130747	H -0.6348968	-2.5765825	5.7403275	H 3.0149114	-5.3643891	-0.6075029

H	4.2514598	-5.2240593	1.5249284	C	-0.4094463	7.3661454	-2.1793393	C	-0.4770424	2.0426037	0.3000452
H	3.2498093	-4.1324687	3.4988770	C	1.7903035	7.0708631	-1.4275572	C	1.1963954	1.1829498	-1.1961089
N	0.8404326	-2.9726842	3.5454905	H	1.0396913	5.8789458	0.3557012	H	1.7625446	0.3398852	-1.5707817
N	0.5756259	-4.3506219	-1.0278267	H	2.7982860	6.6717841	-1.5273003	C	2.5472956	2.6035233	-2.7274277
N	-3.3446155	-2.8032974	1.2353062	H	1.8036216	8.0725182	-1.0031535	H	-0.7652839	4.1514835	0.2292815
H	-4.3483933	-2.8206550	1.2995923	C	0.8718669	4.7838203	-4.1256252	H	-1.2197137	1.8793297	1.0678276
H	0.9607517	-4.9145682	-1.7666791	N	0.7937828	4.2809286	-5.1570715	C	2.1558548	2.4738581	-3.7366770
H	1.3179713	-3.0511334	4.4274931	C	-0.0232352	-0.3705807	0.2531119	H	3.3158407	1.8496040	-2.5663322
C	0.9680124	6.9793596	-2.7339705	H	1.3146740	7.5238232	-3.6054905	H	3.0121400	3.5871838	-2.6845659
C	0.9649280	5.4453064	-2.8762942	C	1.4605920	2.4533002	-1.6987746	H	-1.1656789	7.8992154	-2.7341704
C	0.8816211	4.9176407	-1.6438303	C	0.7241607	3.5368517	-1.1888772	H	-1.3355569	6.8786681	-0.2819695
C	0.8239345	6.1198949	-0.6802712	C	-0.2146732	3.3100529	-0.1731754				
C	-0.4940389	6.8555527	-0.9575329	C	0.2309185	0.9542328	-0.2180916				

2b Quadricyclane-ethinyl-TATA

E_{M062x-D3/def2TZVPP} = -1605.149206215

Nmag = 2 (-20.93 cm⁻¹; -9.38 cm⁻¹)

C	-0.4920794	-1.4529178	0.5637063	C	2.3648607	-4.6692074	-0.3724295	H	1.5803935	8.1676805	-1.1365444
C	-0.6497178	-2.8474594	1.0339735	C	3.2503876	-4.5512686	0.6874022	C	1.8153984	4.4849070	-3.9889471
C	0.6905283	-3.5411787	0.9486421	C	2.8711554	-3.9696366	1.8874712	N	2.2584441	3.7623320	-4.7646901
C	-1.1341775	-2.8431952	2.4661788	C	1.5764238	-3.4681948	2.0196116	C	-0.3406944	-0.3260744	0.1716934
C	-1.6521433	-3.5606216	0.1557829	H	2.6662243	-5.1455667	-1.2959930	H	2.3854789	7.2403303	-3.5344409
C	-1.2252173	-4.1886762	-1.0113497	H	4.2552340	-4.9375438	0.5814181	H	-0.1472328	6.8140257	-4.1701537
C	-2.1615047	-4.7197010	-1.8983923	H	3.5650125	-3.9040339	2.7151366	H	-1.4960300	6.1783506	-1.7529481
C	-3.5100246	-4.6290828	-1.5908788	N	1.1413604	-2.8839265	3.2047149	C	1.3828491	2.8335390	-0.8097147
C	-3.9465520	-4.0463391	-0.4111764	N	0.1395976	-4.2686432	-1.2623338	C	0.2912287	3.6098636	-1.2288832
C	-3.0068563	-3.5154283	0.4722432	N	-3.3908861	-2.9311621	1.6751103	C	-0.9923770	3.0768596	-1.1789070
C	-0.2122939	-2.7884330	3.5081096	H	-4.3640119	-3.0099840	1.9180068	C	-0.1392530	1.0086375	-0.3004507
C	-2.4994852	-2.8126794	2.7361005	H	0.4174890	-4.7994281	-2.0704374	C	-1.2157592	1.7910799	-0.7147000
C	-2.9466326	-2.6632881	4.0488097	H	1.7711986	-2.9396377	3.9871547	C	1.1510065	1.5448698	-0.3551056
C	-2.0164256	-2.5644119	5.0717509	C	1.7019674	6.8108616	-2.8178765	H	1.9824647	0.9340598	-0.0279695
C	-0.6548805	-2.6381868	4.8222882	C	1.2703154	5.3728702	-3.0219623	C	2.7767330	3.3963533	-0.8425132
H	-4.0073038	-2.6289898	4.2599392	C	0.5161198	4.9838844	-1.7199147	H	-1.8260071	3.6794573	-1.5156086
H	-2.3616269	-2.4451470	6.0899393	C	0.6297525	6.2324134	-0.8755158	H	-2.2159864	1.3829082	-0.6769466
H	0.0604890	-2.5835176	5.6323330	C	-0.4356858	6.1274493	-1.9385272	H	3.1174578	3.5506670	-1.8680076
H	-5.0008288	-4.0033645	-0.1714096	C	0.2938146	6.5132441	-3.2347611	H	3.4785479	2.7231449	-0.3551211
H	-1.8328189	-5.1984179	-2.8113169	C	1.7652195	7.1124688	-1.3402804	H	2.8160902	4.3620323	-0.3355287
H	-4.2372912	-5.0393975	-2.2786082	H	0.3357150	6.1603382	0.1610384				
C	0.0710642	-4.1663107	-0.2357381	H	2.7217615	6.8264237	-0.9024730				

3a Norbornadiene-TOTA

E_{M062x-D3/def2TZVPP} = -1318.251824257

Nmag = 0

C	0.0303869	-0.3511217	0.0093937	H	-0.2781967	-0.0107475	4.6261669	C	-0.2121972	3.1072510	0.8141606
C	1.3420349	-1.0221292	0.2740226	H	-3.9151937	-1.7686390	-1.9316991	C	-0.6039523	3.7295868	-0.2906174
C	-0.7457136	-0.3070325	1.2894456	H	-0.2096073	-3.0428432	-3.7398128	C	1.5817329	2.9810505	-0.6931381
C	-0.7297440	-1.1668789	-0.9776950	H	-2.6892449	-2.9282005	-3.7525818	H	1.5267044	1.7560412	1.2190608
C	-0.0526605	-1.8771252	-1.9544285	C	1.9566344	-1.7459952	-0.7380835	H	2.3195066	2.4244061	-1.2685206
C	-0.7483791	-2.5153255	-2.9666958	C	3.2168738	-2.2907561	-0.5446882	H	1.9969285	3.9128835	-0.3156694
C	-2.1361776	-2.4470080	-2.9580989	C	3.8271663	-2.1296525	0.6931762	C	-0.8639888	1.2073038	-2.7175994
C	-2.8358062	-1.7987171	-1.9477594	C	3.1928158	-1.4785037	1.7444906	N	-1.4372901	0.9026812	-3.6666912
C	-2.1144738	-1.1678017	-0.9497075	C	1.9342613	-0.9413044	1.5248404	H	0.2450549	3.7415778	-2.3689055
C	-0.0882837	-0.2360257	2.5078261	C	3.6902777	-2.8390446	-1.3460414	H	-1.4054541	4.4419521	-0.4081380
C	-2.1323889	-0.3281868	1.2480166	H	4.8098448	-2.5512750	0.8532533	H	-0.6177832	3.1861946	1.8113661
C	-2.8704806	-0.1641679	2.4103500	H	3.6472956	-1.3961991	2.7209928	O	1.2824772	-0.3324287	2.5730844
C	-2.1904889	-0.0186140	3.6133187	C	0.2441183	3.1763812	-1.4434643	O	1.3227328	-1.9451889	-1.9370433
C	-0.8033025	-0.0741643	3.6843094	C	-0.1432466	1.6879500	-1.5909415	O	-2.7904529	-0.5231175	0.0621837
H	-3.9495247	-0.1672460	2.3630357	C	0.2480325	1.0660811	-0.4730302				
H	-2.7577960	0.1083719	4.5251978	C	0.9069482	2.1357278	0.4133587				

3b Quadricyclane-TOTA

E_{M062x-D3/def2TZVPP} = -1318.226393475

Nmag = 0

C	0.0061829	-0.2663180	0.1019546	C	-1.0856375	-0.2002219	3.7100225	H	4.7372488	-2.4523368	1.1912893
C	1.2952103	-0.9497219	0.4240347	H	-4.1321166	-0.4116404	2.1839295	H	3.4391516	-1.3622045	3.0073400
C	-0.8673625	-0.3232148	1.3141530	H	-3.0974024	-0.1590752	4.4287937	O	1.0759732	-0.3068098	2.7199027
C	-0.6628669	-1.0177742	-1.0034119	H	-0.6238301	-0.1446450	4.6850072	O	1.4751725	-1.7192327	-1.8361984
C	0.1051800	-1.6523562	-1.9689094	H	-3.7468299	-1.7257810	-2.1963387	O	-2.8247284	-0.6292435	-0.0395596
C	-0.4891891	-2.2507742	-3.0667525	H	0.1249975	-2.7274976	-3.8165603	C	-0.0171503	3.3606456	-1.3172645
C	-1.8772693	-2.2367703	-3.1552149	H	-2.3541478	-2.6964228	-4.0101575	C	-0.7139158	2.3431798	-0.4315860
C	-2.6681507	-1.6908088	-2.1538345	C	1.9987600	-1.6096470	-0.5709343	C	0.2990561	1.1706492	-0.3030867
C	-2.0459605	-1.1015356	-1.0619001	C	3.2481389	-2.1504571	-0.3092725	C	1.4564655	1.6131371	-1.1602513
C	-0.2948222	-0.2693854	2.5765276	C	3.7637243	-2.0336771	0.9764523	C	1.3228780	2.1242525	0.2541833
C	-2.2430847	-0.4371384	1.1872189	C	3.0509072	-1.4226615	2.0010432	C	1.0279293	2.6658945	-2.1535188
C	-3.0602172	-0.3620850	2.3051467	C	1.8019465	-0.8927239	1.7130851	H	2.2380206	0.9017713	-1.3801845
C	-2.4677453	-0.2226779	3.5521713	H	3.7898972	-2.6541903	-1.0965201	C	-2.1144788	2.3768862	-0.1845161

N -3.2436121 2.4471762 0.0161454
C 0.3298704 3.2887176 0.1368149
H -0.6051590 4.1896293 -1.6807240
H 0.6067968 2.2307517 -3.0597319
H 1.8444056 3.3368115 -2.4216740
H 0.1556229 4.0733537 0.8535434
H 2.0421408 1.9561725 1.0393432

References

- [1] Gunes, Y.; Arcelik, N.; Sahin, E.; Fleming, F. F.; Altundas, R. *Eur. J. Org. Chem.* **2015**, 6679-6686.
- [2] Laursen, B. W.; Krebs, F. C. *Chem. Eur. J.* **2001**, 7, 1773-1783.
- [3] Browne, D. L.; Baumann, M.; Harji, B. H.; Baxendale, I. R.; Ley, S. V. *Org. Lett.* **2011**, 13, 3312-3315.
- [4] Martin, J. C.; Smith, R. G. *J. Am. Chem. Soc.* **1964**, 11, 2252-2256.
- [5] Zhao, Y.; Truhlar, D. G., *Theor. Chem. Account* **2008**, 120, 215-241.
- [6] Grimme, S.; Antony, J.; Ehrlich, S.; Krieg, H., *J. Chem. Phys.* **2010**, 132, 154104.
- [7] Weigend, F.; Häser, M.; Patzelt, H.; Ahlrichs, R., *Chem. Phys. Lett.* **1998**, 294, 143.
- [8] Goerigk, L.; Hansen, A.; Bauer, C.; Ehrlich, S.; Najibi, A.; Grimme, S., *Phys. Chem. Chem. Phys.* **2017**, 19, 32184-32215.
- [9] Turbomole7.2: TURBOMOLE V7.2 2017, a development of University of Karlsruhe and Forschungszentrum Karlsruhe GmbH, 1989-2007, TURBOMOLE GmbH, since 2007; available from <http://www.turbomole.com>.